POWDER INJECTION MOULDING OF SYNTHESIZED HYDROXYAPATITE POWDER FROM CLAMSHELL

NURUL HUDA BINTI M. ALI

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Faculty of Chemical Engineering

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CONFIRMATION BY PANEL OF EXAMINERS

I certify that a Panel of Examiners has met on 28th April 2015 to conduct the final examination of Nurul Huda Binti M. Ali on her Master of Science thesis entitled "Powder Injection Moulding of Synthesized Hydroxyapatite Powder from Clamshell" in accordance with Universiti Teknologi MARA Act 1976 (Akta 173). The Panel of Examiners recommends that the student be awarded the relevant degree. The Panel of Examiners was as follows:

Jailani Salihon, Ir. PhD
Professor
Faculty of Chemical Engineering
Universiti Teknologi MARA
(Chairman)

Norliza Ibrahim, PhD
Senior Lecturer
Faculty of Chemical Engineering
Universiti Teknologi MARA
(Internal Examiner)

Abu Bakar Sulong, Ir. PhD
Associate Professor
Faculty of Mechanical and Materials Engineering
Universiti Kebangsaan Malaysia
(External Examiner)

SITI HALIJJAH SHARIFF, PhD
Associate Professor
Dean
Institute of Graduates Studies
Universiti Teknologi MARA
Date: 17th August, 2015
ABSTRACT

Hydroxyapatite (HAp) is one of the most versatile bioceramic materials since it is widely used in biomedical and dentistry applications. This research focused on the effect of pH and calcination temperature on synthesized HAp powder from clamshells via chemical precipitation method. Besides that, the process of PIM using synthesized HAp powder mixed with PS based binder system had studied. The effect of sintering temperatures on the as-sintered HAp specimens had evaluated for its physical and mechanical properties. The process to synthesize HAp powder involves calcination of clamshells powder followed by the titration method using low concentration of phosphoric acid. Based on the analysis, synthesized HAp powder at the temperature of 850°C with the final pH solution of 6.5 has similar characterizations with commercial HAp powder. Then, the process is followed by a green processing route via PIM technique to produce the as-sintered HAp specimen. From the results attained, it shows that synthesized HAp feedstock prepared with the powder loading of 60 vol.% shows pseudo-plastic behavior. The feedstock was successfully injection moulded according to ASTM standard C1424-10 to produce green specimens at the temperature of 200°C with the pressure range from 4 to 7 bars. The green specimens were then successfully debound and sintered through single step wick-debinding and sintering process using alumina powder as an embedment agent. The sintered specimens were than evaluated on physical and mechanical properties. From the results obtained, sintering temperature above 1100°C is not preferable since HAp is start to decompose and forming TCP which not suitable for load bearing applications.
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CHAPTER ONE
INTRODUCTION

1.1 RESEARCH BACKGROUND

Over the past of several years, owing to greater demands of implant material in biomedical sectors, it seems that research on hydroxyapatite [HAp, Ca_{10}(PO_{4})_{6}(OH)_{2}] powder has addressed remarkable attention by many materials researchers and medical practitioners. The main reason of HAp is widely used as an implant material (Prabakaran and Rajeswari, 2009) is due to its chemical similarity in natural bone material with the composition of 70% in human bone (Cui, Nelson, Peng, et al. 2012). In addition, HAp contains a compound of calcium (Ca) and phosphorus (P) with the molar ratio of Ca/P approximately equal to 1.67 which is almost equal to the Ca/P ratio for the natural bone (Mittal, Prakash, Nath and Sapra, 2010). Thus, it makes HAp becomes a suitable material for the implant purposes.

It is well known that HAp has been used extensively in various biomedical applications such as in bone repair, orthopedic, as coating for metallic prostheses and bone grafting applications (Maclaine, Meek and Brydone, 2010), (Afshar, Ghorbani, Ehsani, Saeria and Sorrell, 2003). This is due to its excellent biocompatibility, bioactivity, osteoconductivity and non-toxicity properties (Kumar, Thamizhavel and Girija, 2012). However, the critical issue in utilizing the HAp is its cost tremendously expensive. Therefore, researchers around the world have been attempting to find an alternative route in producing HAp by synthesis technique from any potential waste material using the concept of “waste to wealth”.

Recently, HAp has been successfully synthesized from natural waste material sources such as oyster shells (Wu, Hsu, Wu and Ho, 2011), corals (Ripamonti, Crooks, Khoali and Roden, 2009) eggshells (Prabakaran and Rajeswari, 2009) (Kumar et al., 2012) and seashell (Vecchio, Zhang, Massie, Wang and Kim, 2007) owing to high content of CaCO_3 which is the important calcium source for the preparation of HAp (Ho, Hsu, Hsu, Hung and Wu, 2013). However, the research on HAp from waste material of