INVESTIGATION ON THE ROLE OF TANNIC ACID AS CORROSION INHIBITOR FOR ALUMINIUM ALLOY IN CHLORIDE SOLUTION VIA WEIGHT LOSS TEST

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LIST OF TABLES

Table	Caption	Page
1.0	Hydrochloric acid resistant metal materials	20
3.1	Concentrations of corrosion inhibitors prepared	22
3.2	Chemical composition of aluminium alloy 1100 coupon	23
3.3	List of chemicals that will be used in the study	23
3.4	List of instruments that will be used in study	24
3.5	List of equipments used in study	24
4.1	Characteristics of peak bands of FTIR spectrum for corrosion inhibitor	29
4.2	Calculated value of corrosion rate and inhibition efficiency for 0.1 M HCl in the presence of tannic acid as corrosion inhibitor from weight loss of aluminium alloy 1100	30
4.3	Calculated value of corrosion rate and inhibition efficiency for 0.5 M HCl in the presence of tannic acid as corrosion inhibitor from weight loss of aluminium alloy 1100	31
4.4	Calculated value of corrosion rate and inhibition efficiency for 1.0 M HCl in the presence of tannic acid as corrosion inhibitor from weight loss of aluminium alloy 1100	31

ABSTRACT

INVESTIGATION ON THE ROLE OF TANNIC ACID AS CORROSION INHIBITOR FOR ALUMINIUM ALLOY IN CHLORIDE SOLUTION VIA WEIGHT LOSS TEST

In this study, the role of tannic acid was investigated as corrosion inhibitor for aluminium alloy 1100 in chloride solution through weight loss test. The main goals of this study are to determine the corrosion rate and corrosion inhibition efficiency of alloys and to investigate the effect of localised corrosion on aluminium surface by using optical microscope. In this work, tannic acid standard has been used as a corrosion inhibitor and was subjected to three hours of immersion test in 0.1 M, 0.5 M and 1.0 M of HCl. As the results of the study, it clearly shows that as the concentration of inhibitors increases, more inhibitor molecules are adsorbed onto the surface of the tested coupon. Thus, corrosion inhibition efficiency improves. In this study, it shows that the highest inhibition efficiency is at 98.73% with the concentration of 3 g/L tannic acid where the concentration of HCl is 0.1 M and lowest inhibition efficiency is at 96.17% with 4 g/L of tannic acid. The aggressive chloride ion that corrode the surface of the aluminum alloy caused scratches and little brittles to form on aluminium alloy's surface. This corrosion resulted in a dissolving process known as localized corrosion. As a conclusion, the produced ferric tannate worked as a thin protective layer or film that possibly coated the aluminium alloy 1100, and inhibit further corrosion. The outcomes of this study will help to design long-term corrosion protection solutions for aluminium alloys 1100 in chloride environments.

Keywords: tannic acid, corrosion inhibitor, aluminum alloy, chloride solution, weight loss analysis