# ELECTRONIC PROPERTIES OF LiNbO3 USING DENSITY FUNCTIONAL THEORY

L.W. Zainuddin<sup>1,2\*</sup>, O.H. Hassan<sup>3</sup>, M.F.M. Taib<sup>2,3</sup>

<sup>1</sup>Faculty of Applied Sciences, Universiti Teknologi MARA, Perak Branch Tapah Campus, 35400 Tapah Road, Perak, Malaysia

<sup>2</sup>Faculty of Applied Sciences, Universiti Teknologi MARA, 40450 Shah Alam, Selangor, Malaysia

<sup>3</sup>Ionic Materials and Devices (iMADE), Institute of Science, Universiti Teknologi MARA, 40450 Shah Alam, Selangor, Malaysia.

#### \*liliwidarti@uitm.edu.my

**Abstract:** Using the density functional theory, the structural and electronic properties of  $LiNbO_3$  are investigated. The lattice constants and unit cell volume were calculated from the optimized unit cell, which were in well agreement with the reported theoretical as well as experimental values. Electronic band structure and density of states (DOS) demonstrated its semiconducting nature showing a band gap of 3.504 eV.

Keywords: LiNbO<sub>3</sub>, DFT, electronic properties.

### INTRODUCTION

LiNbO<sub>3</sub> were firstly synthesised in Bell Laboratories in 1966. It is an artificial dielectric substance that does not occur in nature. It was found to be ferroelectric for the first time in 1949 (Weis & Gaylord, 1985). Ferroelectric lithium niobate (LiNbO<sub>3</sub>) is widely used in the field of integrated and guided-wave optics due to its advantageous optical, piezoelectric, electro-optic, elastic, photoelastic, and photorefractive characteristics(Sanna & Schmidt, 2017). The common synthesized method in producing LiNbO<sub>3</sub> crystal are Czochralski method, sol-gel method, citrate method and a few more method (Masloboeva et al., 2020)(Reza & Fray, 2014).

LiNbO<sub>3</sub> has gain researcher attention due to its unique properties. LiNbO<sub>3</sub> belongs to a trigonal crystal system and can be describe as hexagonal or rhombohedral primitive unit cell. This material exhibit in two different phases according to the temperature. LiNbO<sub>3</sub> exhibits a ferroelectric phase at room temperature with a space group of R3c and possess a paraelectric phase with a space group of R3c above 1480K (Dan-dan et al., 2015)(Saleev et al., 2018). LiNbO<sub>3</sub> also owns diverse physical properties such as pyroelectric, piezoelectric, photo-elastic, electrooptic and large birefringence (Sanna & Schmidt, 2012). In addition, LiNbO<sub>3</sub> has a high Curie temperature of 1480 K with a high melting point. It also exhibit a wide band gap energy of 3.78 eV (Thierfelder et al., 2010) and a large spontaneous polarization about 0.70 C/m<sup>2</sup> (Sanna & Schmidt, 2012).

In the last 20 years, computational methods have become an important and necessary part of material science study and it can give a high accuracy detailed feature of materials, such as their structural, electronic, and optical properties. Kohiki et al. conducted a comparative investigation of the energy loss function using theoretical calculation and experimental measurement with X-ray photoemission spectra (Kohiki et al., 1998). Comparative structural and physical studies of the paraelectric and ferroelectric phases of LiNbO<sub>3</sub> are also available. Numerous studies on the structural, electronic, and optical properties of LiNbO<sub>3</sub> were conducted using theory (Husin et al., 2019)(L. W. Zainuddin et al., 2022)

In this paper, the structural and electronic properties of  $LiNbO_3$  have been performed using density functional theory. This study is carried out using Cambridge Serial Total Energy Package (CASTEP) computer code within the exchange correlation of local density approximation (LDA) approach. The objective of this work is to investigate the density of state and band gap energy of LiNbO<sub>3</sub>.

### METHODOLOGY

The ground state properties of any material can be computed using the density functional theory (DFT). In this paper, the plane wave DFT is used to optimize the geometry and calculate the desired properties of LiNbO<sub>3</sub>. The first-principles calculations performed in this work are based on density functional theory (DFT) within exchange-correlation functional from Local Density Approximation (LDA) using Cambridge Serial Total Energy Package (CASTEP) computer code. Geometrical optimization is first performed using the LDA method on LiNbO<sub>3</sub> and followed by its electronic properties. In LDA functional, the cutoff energy used in this work is 380 eV with k-point

grid of  $3 \times 3 \times 2$ . Geometrical optimization was set at 5.0 x  $10^{-6}$  eV/atom for total energy, 0.01 eV/Å for maximum force, 0.02 GPa for maximum stress and 5.0 x  $10^{-4}$  Å for maximum displacement.

#### **RESULTS AND DISCUSSION**

This section analyzes the structural and electronic properties of LiNbO<sub>3</sub> crystal in details. Table 1 shows the lattice parameter for LiNbO<sub>3</sub> via experimental and theoretical values. From Table 1, the optimized lattice parameter as compared to the experimental values shows an underestimated value of 1.8% for lattice and b, 1.4% for lattice c and 4.7% for cell volume. As comparaison with others theoretival results, the lattice parameter shows a comparable and acceptable values. Figure 1 shows the band structure and partial density of states (DOS) fro LiNbO<sub>3</sub>. From the band structure, the band gap energy of 3.504 eV is observed. This value indicate that LiNbO<sub>3</sub> has wide band gap energy. In the same figure, the density of state of LiNbO<sub>3</sub>. At valance band (VB) from -6 eV to 0 eV was mainly derived from O 2p state meanwhile from 0 eV to 3 eV shows a gap which indicate the band gap energy similar to the band structure. From 3 eV to 6 eV, the main peak were corespond to Nb 4d state followed by O 2p state and Li 2s state. It clearly observed that there is overlapping between Nb 4d and O 2p states which show a strong covalent bonding between Nb and O.

Table 1. Structural Parameters of LiNbO <sub>3</sub> from Theoretical and Experimental				
a (Å)	b (Å)	c (Å)	Volume (Å <sup>3</sup> )	References
5.058	5.058	13.699	303.631	This work
5.057	5.057	13.942	308.58	Theoretical (Hossain, 2019)
5.186	5.186	14.009	326.289	Theoretical (Boukhtouta et al., 2020)
5.150	5.150	13.864	318.460	Experimental (Redfield & Burke, 1974)



Figure 1. Band structure and density of states (DOS) of LiNbO3 using LDA.

## DIFFERENCE CONTENT OF CALCIUM IN SUFFY PURE GOAT MILK AND COW MILK - A PRELIMINARY STUDY

Nunshaimah Salleh, Adibatul Husna Fadzil, Norsakina Zurina Zulkifli, Rosliza Ali, Yanti Yaacob, Nordiana Suhada Mohmad Tahirudin.

## GOAT MILK SOAP FROM EXPIRED POWDERED MILK

Rosliza Ali, Norsakina Zurina Zulkifli, Nunshaimah Salleh, Adibatul Husna Fadzil, Yanti Yaacob, Nordiana Suhada Mohmad Tahirudin.

33

Pejabat Perpustakaan Librarian Office

Universiti Teknologi MARA Cawangan Perak Kampus Seri Iskandar 32610 Bandar Baru Seri Iskandar, Perak Darul Ridzuan, MALAYSIA Tel: (+605) 374 2093/2453 Faks: (+605) 374 2299

KNOLIKH

ERIMA

Universiti Teknologi MARA Pe

ABATRE

JAN 2023

Surat kami

OGIA,

:

π



700-KPK (PRP.UP.1/20/1)

20 Januari 2023

Prof. Madya Dr. Nur Hisham Ibrahim Rektor Universiti Teknologi MARA Cawangan Perak

Tuan,

### PERMOHONAN KELULUSAN MEMUAT NAIK PENERBITAN UITM CAWANGAN PERAK **MELALUI REPOSITORI INSTITUSI UITM (IR)**

0

EP

NN

25

Tindakan

Perkara di atas adalah dirujuk.

2. Adalah dimaklumkan bahawa pihak kami ingin memohon kelulusan tuan untuk mengimbas (digitize) dan memuat naik semua jenis penerbitan di bawah UiTM Cawangan Perak melalui Repositori Institusi UiTM, PTAR.

Tujuan permohonan ini adalah bagi membolehkan akses yang lebih meluas oleh pengguna perpustakaan terhadap semua maklumat yang terkandung di dalam penerbitan melalui laman Web PTAR UiTM Cawangan Perak.

Kelulusan daripada pihak tuan dalam perkara ini amat dihargai.

Sekian, terima kasih.

nar

"BERKHIDMAT UNTUK NEGARA"

Saya yang menjalankan amanah,

SITI BASRIYAH SHAIK BAHARUDIN Timbalan Ketua Pustakawan

PROF. MADYA DR. NUR HISHAM IBRAHIM REKTOR UNIVERSITI TEKNOLOGI MARA CAWANGAN PERAK KAMPUS SERI ISKANDAR

Universiti Teknologi MARA Cawangan Perak : Experiential Learning In A Green Environment @ Seri Iskandar

Powered by CamScanner