

**UNIVERSITI TEKNOLOGI MARA**

**CATALYTIC  
TRANSESTERIFICATION OF  
WASTE COOKING OIL INTO  
BIODIESEL USING VARIOUS  
POTASSIUM BASED  $\text{Al}_2\text{O}_3$   
CATALYST**

**MUHAMMAD AMIRRUL HAKIM BIN  
LOKMAN NOLHAKIM**

Thesis submitted in partial fulfillment  
of the requirements for the degree of  
**Master of Science  
(Chemistry)**

**Faculty of Applied Sciences**

**August 2023**

## ABSTRACT

The waste cooking oil, WCO (from household area taman scientex, Pasir Gudang) obtained was found to have low in free fatty acids (<2%) and will undergo a transesterification process directly by using potassium salt as based catalyst. Potassium carbonate ( $K_2CO_3$ ) and potassium hydroxide (KOH) as potassium-based salt precursor was chosen and immobilized onto the surface of  $Al_2O_3$ , the catalyst was named as  $K_2CO_3/Al_2O_3$  and  $KOH/Al_2O_3$ . The catalyst was prepared via incipient wetness impregnation (IWI) method. The perforated hydrophilic materials (PHM) that made from high-density polyethylene (HDPE) was used as the catalyst reactor bag. The prepared catalysts were investigated using TGA, XRD, BET, SEM-EDX, TPD and for leaching of the prepared catalyst was studied by using FTIR, ICP-AES and XRF analysis. The optimum condition for transesterification of WCO were 700°C calcination temperature, 10 wt% catalyst concentration, 5 wt% catalyst loading, 2 h reaction time and 60°C reaction temperature. Based on the surface area for the beads catalyst by using PHM, found to have larger surface area compared to powder catalyst by showing 97.97% biodiesel yield for  $KOH/Al_2O_3$  powder catalyst (2.36 m<sup>2</sup>/g of surface area and 6.24 nm of pore size), 93.58% biodiesel yield for  $KOH/Al_2O_3$  beads catalyst (133.37 m<sup>2</sup>/g of surface area and 7.08 nm of pore size), 88.94% biodiesel yield for  $K_2CO_3/Al_2O_3$  powder catalyst (1.51 m<sup>2</sup>/g of surface area and 5.43 nm of pore size) and 96.70% biodiesel yield for  $K_2CO_3/Al_2O_3$  beads catalyst (199.94 m<sup>2</sup>/g of surface area and 6.51 nm of pore size). In addition, from the reusability study, bead catalyst of  $KOH/Al_2O_3$  and  $K_2CO_3/Al_2O_3$  beads catalyst successfully can be recycle up to 8 and 12 times respectively by using PHM as a reactor container. The final product also undergo evaluation on FAME yield by using GC-FID and the data shows that the result different are ~0.5 to ~1.0% only. The properties of biodiesel also comply with ASTM D6751-2 and EN14142 standard method. Most significantly, this study has explored the viability of  $K_2CO_3$  and KOH supported onto  $Al_2O_3$  have a novel value and can be produced with very low cost. This heterogeneous catalyst is stable and suitable to be used in the industrial scale.

## ACKNOWLEDGEMENT

First of all, Alhamdulillah, all praise and gratitude to my creator, the Almighty Allah SWT for allowing me the opportunity to embark on my Master's and for completing this long and challenging journey successfully. I want to take this opportunity to express my solemn thanks to my supervisor ChM Dr. Norshahidatul Akmar Mohd Shohaimi, and Co-supervisors Assoc. Prof. Dr. Mohd Lokman Ibrahim and Dr. Wan Aini Wan Mokhtar for guiding me throughout this project. Their advice, suggestion, and insight has helped me from the initial phase of this research project and then guided me in achieving the goals of this challenging research.

All my research projects would not have been possible without the financial and support from FRGS Grant. My appreciation goes to say thanks to all the crewmembers of the laboratory assistance from chemistry, biology, and physics, and also my research mates who shared their knowledge and information in preparing my thesis. They also give cooperation and always helpful when conducting lab works in the laboratory. I am grateful to Universiti Teknologi MARA (UiTM) Cawangan Pahang for providing the necessary facilities to complete this work.

Most importantly, this thesis is dedicated to my beloved family for their vision and determination to educate me. Also, who offered me unconditional love, understanding and endless support throughout this research project, my wonderful brother and sister, and all my dearest friends who are always with me when I need them.

# TABLE OF CONTENTS

	<b>Page</b>
<b>CONFIRMATION BY PANEL OF EXAMINERS</b>	<b>ii</b>
<b>AUTHOR'S DECLARATION</b>	<b>iii</b>
<b>ABSTRACT</b>	<b>v</b>
<b>ACKNOWLEDGEMENT</b>	<b>vv</b>
<b>TABLE OF CONTENTS</b>	<b>vivi</b>
<b>LIST OF TABLES</b>	<b>x</b>
<b>LIST OF FIGURES</b>	<b>x</b>
<b>LIST OF SYMBOLS</b>	<b>xixii</b>
<b>LIST OF ABBREVIATIONS</b>	<b>xiv</b>
<b>CHAPTER ONE: INTRODUCTION</b>	<b>1</b>
1.1 Background of study	1
1.2 Biodiesel	1
1.3 Problem Statement	2
1.4 Objectives	4
1.5 Significance of Study	4
1.6 Scope of Research	5
<b>CHAPTER TWO: LITERATURE REVIEW</b>	<b>6</b>
2.1 The Importance of Biodiesel to Community	6
2.2 Biodiesel and Feedstock	6
2.3 Biodiesel Properties	8
2.4 Biodiesel Production Method	10
2.4.1 Esterification Reaction	10
2.4.2 Transesterification Reaction	10
2.4.3 Other Biodiesel Methods	15

# CHAPTER ONE

## INTRODUCTION

### 1.1 Background of study

Cooking oil produced from palm tree or scientifically *Elaeis guineensis* has been used by human since back to 5000 years ago. From crude palm oil (CPO), the oil will undergo certain process until become refined, bleached, and deodorized palm oil (RBDPO) as a final product that we called as cooking oil. Waste cooking oil (WCO) generated everyday using cooking oil after frying protein, carbohydrates, or fibres. WCO usually mix oil with black coloured appearance, high residue, high water content and have pungent smell because of the presence of another compound. The WCO usually has thrown away through sink or drain caused the oil will solidify and clogged the pathway, wastewater cannot pass through and flooded happens. It increases the cost to maintenance the system and will polluted the water. To make used of WCO to the better version, the one way is to convert WCO to biodiesel for combustion fuels. Naturally, biodiesel is using vegetable oil or animal fats and react chemically with the catalyst and alcohol through esterification or transesterification process to produce fatty acids methyl ester (FAME). (Fereidooni & Mehrpooya, 2017; Nang et al., 2009).

Biodiesel is much more eco-friendly and non-toxic towards environment compare to existing diesel (Rashid & Anwar, 2008). The example of oil that been used to produce biodiesel are red palm oil (RPO), sunflower, canola, soybean, palm kernel (PKO), crude palm kernel (CPKO) and WCO (Lokman et al., 2015; Lokman Nohakim et al., 2021).

### 1.2 Biodiesel

Naturally, biodiesel is produced by using vegetable oil or animal fats and react chemically with the catalyst and alcohol through esterification or transesterification process to produce fatty acids methyl ester (FAME). (Fereidooni & Mehrpooya, 2017; Nang et al., 2009). Biodiesel are much more eco-friendly and non-toxic towards