UNIVERSITI TEKNOLOGI MARA

A STUDY ON STRUCTURAL AND ELECTRONIC PROPERTIES OF Ag AND Sb-DOPED IN Ba-SITE OF YBa₂Cu₃O_{7-δ} SUPERCONDUCTOR VIA COMPUTATIONAL METHOD

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ABSTRACT

The structural and electronic properties of $YBa_2Cu_3O_{7-\delta}$ superconductors doped with Ag and Sb have been predicted theoretically. The Ag ionic radius of 126 pm matches the Ba atom, which is 134 pm, promoting electrical conductivity. When an Sb atom with a lower ionic radius of 76 pm is doped into the Ba-site of YBa₂Cu₃O₇₋₈, it undergoes structural modifications. To further understand its interatomic behaviour, the structural and electronic properties of both elements doped at the Ba-site of YBa₂Cu₃O_{7- δ} were computationally analysed, over a wide range of dopant percentages. Focuses were made at $x = \pm 0.200$, where Ag dopant enhanced electrical conductivity, and Sb dopant provided structural alterations from prior experiments. The CASTEP computer code was applied employing various exchange-correlation functionals such as Local Density Approximation and Generalized Gradient Approximation on standard DFT (LDA CA-PZ, GGA-PBE, GGA-PBEsol, and GGA-WC) with the VCA technique and 2×2×1 supercell Hubbard U correction method, DFT+U (LDA CA-PZ+U, GGA-PBE+U, GGA-PBEsol+U, and GGA-WC+U) with U = 10.0 eV. Based on the computational calculation, the GGA-PBEsol and GGA-PBEsol+U give the best-validated results as compared to the referred experimental and theoretical studies. The lattice parameters of all samples differ by less than 5% from the previous experimental investigation. DFT via GGA-PBEsol indicated considerable bond length changes along the Cu(1)-O(1) and Cu(1)-O(4) by +0.62% and -0.38%, respectively. CuO chain behaves as the charge reservoir to the conducting CuO₂ plane. The bond length along Cu(2)-O(2), Cu(2)-O(3), and Cu(2)-O(4) changes to experimental by +0.52%, +0.36%, and -4.65%, respectively. As Ag dopant was added to the sample, the bond length on the CuO chain and CuO₂ plane decreased, indicating contraction due to impurity addition at the Basite. As Sb dopant was added, an increment of bond length was observed along with Cu(2)-O(2) and Cu(2)-O(3) by +2.89% and +23.95%, respectively, showing an increment of bond length upon Sb doping. The electronic band structure, total and partial density of states (DOS), and electron density show that hybridization exists between the Cu 3d and O 2p atoms, verifying the Van Hove Singularity at the Fermi level. Applying GGA-PBEsol, the bandgap of 3.30 eV for pure YBa₂Cu₃O_{7-δ} shows a 17.50% difference compared to other theoretical work. When the DFT+U method was employed via GGA-PBEsol+U, the calculated electronic bandgap for pure, Ag and Sbdoped samples was increased. The DOS on Cu(1) 3d, O(1) 2p, and O(4) 2p atoms along the CuO chain and Cu(2) 3d, O(2) 2p, and O(3) 2p atoms on the CuO₂ plane were increased when doped with Ag atom. When doped with Sb atom, the increment of Cu(1) 3d, O(1) 2p, and O(4) 2p atoms are +132.84%, +136.73%, and +104.33% on CuO chain and along the CuO₂ plane for Cu(2) 3d, O(2) 2p, and O(3) 2p atoms by +96.70%. +103.36%, and +121.01%, respectively. Impurity dopants have the highest concentration on O atoms due to their electronegativity, increasing their electron density. As Ag and Sb dopants were added to the Ba-site of $YBa_2Cu_3O_{7-\delta}$, respectively, the total charge on Cu(2) increased substantially where Cu atoms exhibit the most atomic population on orbitals. This work will help researchers understand the structural and electronic properties gained by computational methods. This approach could be used in future experiments to establish the Ag and Sb dopant percentages in YBa₂Cu₃O₇₋₈ superconductor, resulting in high-performance sample preparation.

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TABLE OF CONTENTS

		Page
CON	FIRMATION BY PANEL OF EXAMINERS	ii
AUT	THOR'S DECLARATION	iii
ABS	TRACT	iv
ACK	KNOWLEDGEMENT	V
TAB	BLE OF CONTENTS	vi
LIST	Γ OF TABLES	ix
LIST	Γ OF FIGURES	XV
LIST	r of symbols	xxiv
LIST	Γ OF ABBREVIATIONS	XXV
CHA	APTER ONE INTRODUCTION	1
1.1	Research Background	1
1.2	Problems Identification	4
1.3	Objectives	7
1.4	Significance of Study	8
1.5	Scope and Limitations in Study	8
1.6	Thesis Outline	9
CHA	APTER TWO LITERATURE REVIEW	10
2.1	Chapter Overview	10
2.2	Introduction to Superconductivity	10
2.3	Computational Analysis	22
2.4	First Principles Study	23

2.5	CASTEP and Material Studio	38
2.6	Computational Analysis on YBCO	40
2.7	Summary of Chapter	51
СНА	PTER THREE RESEARCH METHODOLOGY	53
3.1	Introduction	53
3.2	Computational Analysis - CASTEP and Material Studio (MS)	53
3.3	Computational Analysis of YBa2Cu3O7-8 Superconductor	56
3.4	Summary of Chapter	77
СНА	PTER FOUR STRUCTURAL PROPERTIES OF PURE, Ag	
AND	Sb-DOPED IN Ba-SITE OF YBa2CU3O7-δ	78
4.1	Introduction	78
4.2	Detail of Calculation	78
4.3	Convergence Test with Respect to Cut-Off Energy	80
4.4	Convergence Test with Respect to k-Points	83
4.5	Lattice Parameters	85
4.6	Orthorhombicity of Ag and Sb-doped YBa _{2-x} R _x Cu ₃ O _{7-δ} , e	108
4.7	Oxygen content $(7 - \delta)$	111
4.8	Bond Length	114
4.9	Validation of Structural Properties Obtained via	
	Computational and Experimental Method	145
4.10	Summary of Chapter	146
СНА	PTER FIVE ELECTRONIC PROPERTIES OF PURE, Ag	
AND	Sb-DOPED IN Ba-SITE OF YBa2CU3O7-δ	148
5.1	Introduction	148
	vii	