

**UNIVERSITI TEKNOLOGI MARA**

**FIRST-PRINCIPLES STUDY ON  
STRUCTURAL, ELECTRONIC AND  
OPTICAL PROPERTIES OF Ln-  
DOPED Bi<sub>2</sub>O<sub>3</sub> (Ln = Eu, Gd AND Er)  
AS A PHOTOCATALYST**

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**MSc**

**August 2021**

## AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the results of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

I, hereby, acknowledge that I have been supplied with the Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.


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## ABSTRACT

Crystal structures of  $\alpha$ - $\text{Bi}_2\text{O}_3$  and  $\beta$ - $\text{Bi}_2\text{O}_3$  were calculated using Cambridge serial total energy package (CASTEP) based on the first-principles plane-wave ultrasoft pseudopotential method within local density approximation (LDA) and generalized gradient approximation (GGA) together with Perdew–Burke–Ernzerhof (GGA-PBE) and Perdew–Burke–Ernzerhof revised for solid (GGA-PBEsol). The structural parameter of  $\alpha$ - $\text{Bi}_2\text{O}_3$  and  $\beta$ - $\text{Bi}_2\text{O}_3$  are in good agreement with previous experimental and theoretical data (Matsumoto, Koyama, Tanaka 2010). The DFT+U method gives the correction due to the underestimate results from DFT as compared to the experimental value. All of the polymorphs were calculated for the total density of states (TDOS) and the partial density of states (PDOS) of Bi, O atoms. Density of states exhibits hybridization of Bi 6s and O 2p orbitals and the calculated charge density exhibit the ionic character in the chemical bonding of this compound. The narrowed band gap ( $E_g$ ) and red shift of light absorption edge are responsible for the photocatalytic activity of  $\beta$ - $\text{Bi}_2\text{O}_3$  for water splitting application.  $\beta$ - $\text{Bi}_2\text{O}_3$  is the best structure among these polymorphs for the photocatalyst application based on the calculated optical properties such as optical absorption and electron energy loss function. Hence,  $\beta$ - $\text{Bi}_2\text{O}_3$  polymorph was used in the modification of  $\text{Bi}_2\text{O}_3$  by doping method. Lanthanide (Ln) elements; Europium (Eu), Gadolinium (Gd) and Erbium (Er) as Ln dopants were determined. The Eu- and Gd-doped  $\text{Bi}_2\text{O}_3$  were observed to have lower band gaps compared to pure  $\text{Bi}_2\text{O}_3$ . The band gap of Er-  $\text{Bi}_2\text{O}_3$  was slightly higher than pure  $\text{Bi}_2\text{O}_3$  but it can be related to the results of the absorption coefficient. The presence of Ln 4f states in Ln-doped  $\text{Bi}_2\text{O}_3$  can be seen from the Density of States (DOS) which explained the narrowing of the band gap. Among the Ln-doped  $\text{Bi}_2\text{O}_3$  (Ln = Eu, Gd and Er), the shifting of light towards a longer wavelength spectrum is obtained from the Gd-doped  $\text{Bi}_2\text{O}_3$ . Overall, the first-principles study in this work from the deepest level of atomic-scale can explain the law of physics from the properties of studied materials and improve the understanding of  $\text{Bi}_2\text{O}_3$  semiconductor as a photocatalyst.

## ACKNOWLEDGEMENT



Alhamdulillah, first and foremost I would like to express my gratitude to Allah S.W.T. for His continuous blessing, giving me strength and ability to complete my study. I would like to thanks Universiti Teknologi MARA for giving me the opportunity to study here.

I would like to thanks my supervisors, Prof. Dr. Ab Malik Marwan Bin Ali and also Dr. Mohamad Fariz Bin Mohamad Taib for their support and motivation along my Master's research journey. Their guidance and advice help me a lot in the difficulties that I faced during my research journey and thesis writing.

Besides, I would like to thanks to the fellow members of Computational and Theoretical Physics and also Ionic Materials and Devices (iMADE) Laboratory, UiTM; Hazrie, Kak Ani, Kak Anum, Mimi, Hafiz, Syazwan, Haziq and other colleagues for a very pleasant and enjoyable working environment. Thank you for the comments and suggestions given as it helps me a lot in my research study.

Not to be forgotten, a big thanks to my husband; Mohammad Usamah Bin Mohammad Subhi Hassan Abdel Nabi, my parents; Fairozie Ashikin Binti Yunus Mohamad and Azhar Bin Othman, my parents in law; Mariani Binti Osman and Mohammad Subhi Hassan Abdel Nabi, and also my siblings; Nurul Syazwani and Ahmad Faiz, for being my pillar of strength that help me a lot through thick and thin during my journey. Without them, I will never be able to do this.

Last but not least, I want to thanks those who always support me directly or indirectly throughout my Master's journey until it becomes a success. And also, I wish to express my sincere gratitude for the financial support that I gain from UiTM and other research grants from the Ministry of Education Malaysia (MOE).

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