

UNIVERSITI TEKNOLOGI MARA

**ELECTROCHEMICAL
PERFORMANCE OF LAYERED
 $\text{LiCo}_{0.9}\text{X}_{0.1}\text{O}_2$ (X = Cr, Mn, Fe, Ni)
CATHODE MATERIALS FOR
LITHIUM-ION BATTERIES**

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MSc

February 2021

AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the results of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

I, hereby, acknowledge that I have been supplied with the Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.

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LiCo_{0.9}X_{0.1}O₂ (X = Cr, Mn, Fe, Ni) Cathode Materials
for Lithium-Ion Batteries

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ABSTRACT

LiCoO₂ is one of the main cathode materials used in commercial lithium-ion batteries due to the high rate capacity, high energy density and good life cycle. However, the LiCoO₂ cathode material is high cost and toxic. Besides, this cathode material also suffers from a few limitations such as low thermal stability and structural degradation when charging at a higher voltage. Therefore, the search for better cathode material is needed to be developed to improve the performance of existing LiCoO₂. One of the approaches is the substitution of LiCoO₂ with transition metal which is cheaper and non-toxic that can enhance the structural stability and electrochemical properties. In this work, LiCo_{0.9}X_{0.1}O₂ (X = Cr, Mn, Fe, Ni) prepared by self-propagating combustion (SPC) method using citric acid as a combustion agent were investigated. The precursors of LiCo_{0.9}X_{0.1}O₂ were annealed at a temperature of 800 °C for 24 h. The materials were characterized using Simultaneous Thermogravimetric Analysis (STA), X-Ray Diffraction (XRD), Field Emission Scanning Electron Microcopy (FESEM), and Energy Dispersive X-Ray Spectroscopy (EDX). All the cathode materials found to be single and pure phase with no impurity peaks detected with the surface morphology of roughly polyhedral-type crystal with some spherical-type. Finally, the electrochemical performance of the materials was studied using discharge cycling in the voltage range of 2.5 V to 4.2 V. Based on the results, the Mn substituted cathode material exhibited a better initial specific discharge capacity of 136.60 mAh/g. Different stoichiometries of LiCo_{1-x}Mn_xO₂ (x = 0.1, 0.2, 0.3, 0.4, 0.5) cathode materials has been further investigated. The initial discharge capacities of all the materials are 136.60 mAh/g, 112.66 mAh/g, 93.75 mAh/g, 66.98 mAh/g, 52.32 mAh/g, respectively. Therefore, only small amount of a Mn content is preferable to enhance the electrochemical performance of LiCoO₂ layered cathode material.

ACKNOWLEDGEMENT

Thanks to Allah Almighty, with His willing giving me the opportunity to complete my research. Upon completion of this project, I would like to express my gratitude to many parties. My deepest gratitude to supervisor, Dr Mohd Sufri Mastuli for his sincere guidance and help for me to complete my Master. I also would like to express my sincere appreciation to my co-supervisor, Madam Kelimah Elong for her sincere guidance and great moral support throughout my research project.

I also would like to thank all the staff of the Centre of Nanomaterials Research (CNR) for their help in utilizing the facilities and instruments. Special thanks to Dr Nurhanna Badar and Ms. Nor Fadilah Chayed for their help and guidance. Lastly, my heartfelt gratitude to all my friends in CNR research group for the motivation and help during the difficult time in completing my research. I also would like to express my gratitude to Prof Dr Zurina Osman from Centre of Ionics University of Malaya for giving permission to access the instrument in her lab.

Finally, a million thanks to my family especially to my beloved mom and dad for the endless support and encouragement made it possible for me to complete this project. I also would like to express my deepest gratitude to my best friends, Syarifah Nabihah and Anas Fahmi for their endless moral support from the beginning until the end.

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