UNIVERSITI TEKNOLOGI MARA

PHYSICOCHEMICAL AND ELECTROCHEMICAL PROPERTIES OF MESOPOROUS SILICA -CARBON ELECTRODE

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MSc

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AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the results of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

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ABSTRACT

Lack of explanation in correlation between materials structure and the electrochemical behaviours becoming a growing concern. In this study, the aims are to synthesise and characterise the physical properties of the mesoporous silica Santa Barbara Amorphous-15 and -16 (SBA-15 and SBA-16) and, followed by fabrication of mesoporous silicacarbon electrode (MCPE) and investigate the electrochemical behaviours. The mesoporous SBA-15 and SBA-16 were synthesized via surfactant templating approach using triblock copolymer P123 and F127 as directing agent and tetraethyl orthosilicate (TEOS) as silica source. The synthesized materials were characterized by X-ray diffraction (XRD), infra-red spectroscopy (IR), thermal gravimetric analysis (TGA), scanning electron microscopy (SEM) and nitrogen absorption-desorption tests. The pore diameters were 5.5 nm and 3.2 nm for SBA-15 and SBA-16 respectively, determined with BJH method based on adsorption data. Three different electrodes were fabricated, carbon paste electron (CPE) and two modified carbon paste electrodes (MCPE): SBA-15/MPCE and SBA-16/MPCE. The fabricated electrodes were tested using cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS). The addition of mesoporous silica to the carbon electrode improved the electrochemical behaviors of the electrode formation of microelectrode thus enhanced migration of the electrons. The SBA-16/MCPE showed better adsorption, response signal and a lower resistance compare to SBA-15/MCPE due to 3D cubic structure of SBA-16 which provide better electron pathway than SBA-15 2D hexagonal structure.

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