EFFECTS OF LICIO₄ AND EC ON THE ENR 50 FOR ELECTROCHEMICAL SYSTEM

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Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of Bachelor of Science (Hons.) Physics In the Faculty of Applied Sciences University Teknologi MARA

Mei 2006

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ACKNOWLEDGEMENTS

Alhamdulillah, praise to Almighty GOD for His bless and mercy for giving me an excellent health to ensure best performance during completing my work especially on this report.

I gratefully appreciate the assistance, support and encouragement of those individuals who has contributed in completed my research studies "Investigates the effects of LiClO₄ and EC on the ENR for electrochemical system". Especially, I would like to recognize the very helpful insights as special appreciate to my supervisor Dr Muhd Zu Azhan Yahya who gives me guidelines how to start this project and showing me the experimental method. I am really appreciated to him that contended to spend a little time for me in consultation.

Also not forgetting En. Abdul Malik Marwan who is my co-supervisor. That giving knowledge and lesson to me between that give some helps in doing my research. Then, I would like to thank to all my classmates and my close friends because of their supporting from behind and comments are important to "Investigates the effects of LiClO₄ and EC on the ENR for electrochemical system" completed on time for the first part. With the good commitments, initiatives, and hard work to learn it also one of a background in doing this proposal.

As a final word, I express my sincere thanks to all of those who has been involved in completing final project. I am really appreciated with all of the knowledge that I got.

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ABSTRACT

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SYSTEM ·

In this study, Epoxidized Natural Rubber (ENR-50) was used as a host polymer. ENR-50, ethylene carbonate (EC) and inorganic lithium perchlorate (LiClO₄) were mixed in the desired proportions and dissolved in 30ml of tetrahydrofuran (THF) solution. The solutions were then poured into various Petri dishes and left to form the film at room temperature. EC was used as a plasticizer. The electrical conductivity of all samples was calculated using the bulk resistance value obtained from the complex impedance plot in the frequency range between 100Hz and 1MHz. The highest electrical conductivity obtained for the salted film containing lithium perchlorate is 8.55×10^{-5} S/cm at room temperature. The plot of conductivity versus wt percent of salt indicates that the conductivity increase with increasing dopant content up to amount of 30wt percent for lithium perchlorate. The highest conductivity value of 1.2×10^{-4} S/cm. The highest conducting film for the plasticized ENR-50 was then used as an electrolyte in the fabrication of lithium-air cell. The discharge capacity obtained from the cell is 574mAhg⁻¹.

INTRODUCTION

1.1 Background

Polymer electrolyte now is widely used as the basis of electrolyte electrochemical device. Polymer are being used increasingly as solid media for substituting the liquid component of electrochemical devices such as sensors, displays, supercapacitors, electrochromic windows and rechargeable batteries. Polymer electrolytes are being favor in industrial of batteries fabrication. This intends most of manufacturer to concentrate on researching in this field. Beside that from this research there are widely develop of the new knowledge. In this study the ENR 50 based polymer electrolyte was prepared with doping LiClO₄ and EC as a plasticizer. Introducing of EC which is as plasticizer in this work is to enhance their conductivity.

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