# EFFECTS OF SALT TYPE ON THE ELECTRICAL PROPERTIES CHITOSAN BASED POLYMER ELECTROLYTE

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#### **ABSTRACT**

## EFFECT OF DIFFERENT SALTS ON THE ELECTRICAL PROPERTIES OF CHITOSAN BASED POLYMER ELECTROLYTE.

Polymer electrolyte based on lithium nitrate, LiNO<sub>3</sub>, lithium triflate, LiCF<sub>3</sub>SO<sub>3</sub> and lithium perclorate, LiClO<sub>4</sub> salt and high molecular weight of chitosan was prepared using solution casting techniques. A film prepared is the highest electrical conductivity and largely amorphous. The ionic conductivity for chitosan-LiNO<sub>3</sub>, chitosan-LiCF<sub>3</sub>SO<sub>3</sub> and chitosan-LiClO<sub>4</sub> were conducted over wide range of frequency and at temperatures between 298 K and 353 K. The conductivity is due to the mobile ions from the salt. The conductivity was calculated using the bulk impedance obtained through impedance spectroscopy using the Cole-Cole plots illustrating the variation of the negative imaginary impedance with the real impedance. Dielectric data were analyzed using the complex permittivity  $e^*$  and dissipative loss,  $\tan \delta$  and relaxation times were determined at various temperatures. The temperature dependent conductivity data obeys Arrhenius relationship.

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