

UNIVERSITI TEKNOLOGI MARA

**SYNTHESIS,
CHARACTERIZATION AND
CATALYTIC APPLICATIONS
OF NITROGEN BASED METAL
ORGANIC FRAMEWORKS
(MOFS)**

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ABSTRACT

Three different group of nitrogen based Metal Organic Frameworks (MOFs) derived from amine, nitro and nitrile group were successfully synthesized using solvothermal of mixed linker method, which are also known as MIXMOF. MIXMOF is formation of "mixed-linker" metal frameworks which consists of two isorecticular ligands (terephthalic acid and nitrogen based carboxylic acid) that randomly combined with metal such as zinc(II) acetate. The ligands syntheses of MOFs were characterized by NMR spectroscopy. The NMR results shows that the methyl group (-CH₃) was successfully removed from the compounds. In addition, the MOF-NH₂, MOF-NO₂ and MOF-N₂ and their palladium(II) complexes were characterized by fourier transform infrared (FTIR), Xray-diffraction (XRD), thermal gravimetric analysis (TGA), and field emission scanning electron microscopy (FESEM). The disappearance of C=O absorption peaks in FTIR spectrum shows the complexation of nitrogen based ligands with metal linker of zinc metal was successful in forming MOFs. Besides that, XRD pattern also shows an attachment of zinc and palladium to the MOFs compounds. Despite of that, the thermal gravimetric analysis revealed that all the MOFs compounds exhibits thermal stability up to 500°C. Instead of characterization, the synthesized MOFs supported palladium complexes were subjected to the carbon-carbon cross coupling reaction, Suzuki and Heck reaction. The results show the Pd/MOF-NH₂ catalyst gave conversion of bromobenzene up to 100% at 100°C using K₂CO₃ for Suzuki reaction and Na₂CO₃ for Heck reaction as a base. These higher activities of Pd/MOF-NH₂ may be explained in terms of functional group of MOFs. The cross coupling Suzuki and Heck reaction using Pd/MOF-NH₂ catalyst shows excellent recyclability of catalyst as well in leaching testing. The catalyst was reused up to five times with little loss of activity. Thus, nitrogen based MOFs had been successfully synthesized and can be applied as an effective supported palladium catalyst for carbon-carbon cross coupling reaction.

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CHAPTER ONE

INTRODUCTION

1.1 RESEARCH BACKGROUND

The first catalyst used in chemical industry was in 1746. However, only pure components were used as catalyst during that time. In year 1900, variations of catalyst were introduced in industry. Since then, catalysis started to impact on the chemical industries and plays a key role in the manufacture of majority of chemicals and materials until now. According to Department of Statistic Malaysia (2015), manufacturing industries profit growth every year. This includes the production of catalyst in chemical industry. In May 2015, the industrial production index (IPI) increased by 4.5% compared in the previous year. Subsequently, the value of product will increase depending on process catalyst such as petroleum products, chemicals, pharmaceuticals, synthetic rubber, plastics and many more. Thus, new catalyst that provides special feature such as higher porosity and reusability is expected to be useful in industry. As a concern to environmental issues, more catalysis activities are relevant to many aspect of environmental science.

Catalyst is defined as a substance that is added to the reaction in order to speed up the rate of the reaction without being consumed or produced in the process. Catalytic reactions have lower activation energy which resulting in higher reaction rate at the same temperature and for the same reactant concentration compare to the uncatalyzed reaction (Davis *et al.*, 2012). According to Haveling (2012), catalyst can be divided into homogeneous and heterogeneous catalyst. Homogeneous catalyst is catalyst that present in same phase as reactants while the heterogeneous catalyst present in different phase as reactants. The biggest problem of homogeneous catalyst is difficulty in catalyst recovery, which led to increasing of production cost. Thus, heterogenization of the catalyst could overcome the recovery problem, lowering the cost and increase their thermal stability (Farnetti *et al.*, 2009).

Since 1994, metal-organic frameworks (MOFs) have emerged as an interesting class of materials for applications in catalysis. Metal organic frameworks are crystalline porous solids which composed of three-dimensional (3D) network of metal