## SORPTION OF AMIDO BLACK 10 B ONTO CALCINED Zn-Al-NO<sub>3</sub> LAYERED DOUBLE HYDROXIDE AND CALCINED Zn-Al-CO<sub>3</sub> LAYERED DOUBLE HYDROXIDE

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#### **TABLE OF CONTENTS**

Page
iii
iv-v
vi
vii-viii
ix
Х
xi

## **CHAPTER 1 INTRODUCTION**

1.1	Background	1
1.2	Problem statement	2
1.3	Significance of study	5
1.4	Objectives of study	6
	Section of Stady	v

# **CHAPTER 2 LITERATURE REVIEW**

2.1	History of hydrotalcite			
2.2	What is Double Layered Hydroxide?			
2.3	Principal of Adsorption			
2.4	The effect of azo dye (Amido Black 10 B)			
	2.4.1 Azo dyes	13		
	2.4.2 Amido Black 10 B	14		
2.5	Previous study	16		

### **CHAPTER 3 METHODOLOGY**

3.1	Chemicals			
3.2	Equipments			
3.3	3.3 Synthesis of Layered Double Hydroxides			
	3.3.1	Calcined Zn-Al-CO <sub>3</sub>	19	
	3.3.2	Calcined Zn-Al-NO <sub>3</sub>	20	
3.4	Adsor	ption Experiment	21	
	3.4.1	Sorption Isotherm	22	
	3.4.2	Adsorption properties calculation	23	
3.5	Charac	cterization of LDHs	24	
	3.5.1	Powder X-Ray Diffractormeter (XRD)	24	
	3.5.2	Fourier Transform Infrared Spectrometry (FTIR)	25	
	3.5.3	UV-Visible Spectrophotometer	25	
	3.5.4	Scanning Electron Microscope	26	
3.6	Optim	ization studies	26	

#### ABSTRACT

## SORPTION OF AMIDO BLACK 10 B ONTO CALCINED Zn-Al-NO<sub>3</sub> LAYERED DOUBLE HYDROXIDE AND CALCINED Zn-Al-CO<sub>3</sub> LAYERED DOUBLE HYDROXIDE

Layered double hydroxides (LDHs) calcined, denoted as CLDHs, have been shown to recover their original layered structure in the presence of appropriate anions. In the light of this so-called "memory effect", the removal of Amido Black (AB), an anionic dye, from aqueous solution by calcined Zn-Al-NO<sub>3</sub> and Zn–Al–CO<sub>3</sub> LDHs was investigated in batch mode. The study looked at the influence of pH values, dye-adsorbent contact time, initial dye concentration, adsorbent dosage, particle size, and various temperatures on the decolorization rate of AB. The adsorption isotherms, described by Freundlich model are L-type. The characterization of the solids CLDHs, both before and after removal of AB, by X-ray diffraction, Scanning Electron Microscope and infrared spectroscopy shows that the AB adsorption on CLDHs is enhanced by reconstruction of a matrix hydrotalcite intercaled by the dye, and the intercalation of the organic ion was clearly evidenced by the net increase in the basal spacing from 7.6 Å for Zn–Al– NO<sub>3</sub> and 7.59 for Zn-Al-CO<sub>3</sub>.