

UNIVERSITI TEKNOLOGI MARA

**REMOVAL OF METALS FROM SIMULATED
WASTE WATER USING PHYSICALLY AND
CHEMICALLY MODIFIED CARBONIZED PALM
SHELL**

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AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the result of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

I, hereby, acknowledge that I have been supplies with Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.

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ABSTRACT

The rapid economic development and technological development today have led to the increase of industrial activity, which causes environmental and ecological ruins. Therefore, they raised a concern on environmental protection due to its emission and uncontrolled pollution. Industrial waste, especially metals will cause harm to the environment and ecosystem if being discharged in high concentration. Accordingly, adsorbent was widely used in metals removal for water treatment from the industrial waste. However, due to the high cost and depletion of coal-based source, the potential of palm shell as low cost adsorbent for the removal of metals element was investigated during this study. The purpose of the present study was to investigate the adsorption efficiency of modified carbonized palm kernel shell in an adsorption column for the removal of Be, Ca, Ca, Co, Cr, Cu, Fe, Li, Mg, Mn, Mo, Ni, Pb, Sb, Sr, Ti, V and Zn ions from aqueous solution. The raw palm shell underwent both the physical and multi solvent treatments. There were 6 types of adsorbent used throughout the study, which are UC1, BC1, BCAB1, BCABC1, BCABCM1 and BCABCM3. The adsorbent was characterized by the surface morphology using FESEM, elemental analyzer and the BET analysis. Surface morphology denoted the development of pores in multi solvent treatment while BET analysis was to identify the adsorbent's surface area, and lastly the elemental analyzer for identifying elemental composition of adsorbents. The results indicated that the removal of metal ions is not similar compared among adsorbents, due to the variety of treatments and the mechanism of hypothetical T-shirt pore formation. For the overall adsorption efficiency, adsorbent derived by the combination of physical, solvent treatment and multiple beds BCABCM3 demonstrated the highest adsorption capacity, with an increase of percentage removal of 92.7% compared to the UC1 (control adsorbent). The adsorbent prepared achieved high metal removal by the hypothetical T-shirt shaped pore, which exhibits high pore volume and high surface area. Moreover, the pore formed from above, below, sides of particles and narrow size in the middle of pores exhibits adsorption and reduces desorption from adsorbent surface.

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CHAPTER ONE

INTRODUCTION

1.1 RESEARCH BACKGROUND

The rapid economic development today has led to the increase of industrial activity, which causes environmental and ecological ruins. Technological development has raised concerns about the environmental protection due to its emission and uncontrolled pollution. Industrial waste produced from industrial activities such as the mills, mines, and factories will cause harm to the environment and ecosystem if being discharged in high concentration. Apart from the industrial waste, the water pollution was also caused by the municipal and agricultural runoff. The municipal water pollution consists of wastewater from homes and commercial establishment while agricultural water pollution consists of manures and slurry.

World Bank (2012) reported that, Malaysia produced 208.3 thousand kg/day of organic water pollutants with 2.8% contributed by primary metals, 16.5% by chemicals, and 6.6% by the textile industry for the year 2007. There were many pollutants contributing to wastewater pollution, including the metals, dyes and from the phenol group. The Organization for Economic Co-operation and Development (OECD), defined heavy metals as potentially toxic metals used in industrial processes, for example, arsenic, cadmium, chromium, copper, lead, mercury, nickel, and zinc, which may damage the plant and animal life at low concentrations and tend to accumulate in the food chain. The significant examples of wastewater pollutions are discharges of wastewater mixed with mercury in Minamata Bay which affects more than 10 000 people around 1950s and 1960s in Japan. Concerning Malaysia, most of the rivers were contaminated and cannot be used as drinking source. According to the Department of Environment (DOE) (2006, cited in BERNAMA, 2006), 10% of the river in Malaysia were heavily contaminated or