

**UNIVERSITI TEKNOLOGI MARA**

**KINETIC ANALYSIS OF PLASTIC  
POLYETHYLENE TEREPHTHALATE (PET)**

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## ABSTRACT

This study provides the effect of temperature on mass change of plastic over time by using Thermogravimetric Analysis (TGA). The main objectives include studying the thermal degradation and kinetic analysis of 20mg plastic Polyethylene terephthalate (PET). Thermal degradation was done by using a Thermogravimetric analyzer with temperature range from 25°C to 800°C at various heating rates. Calculated kinetics parameters obtained by using Ozawa Flynn and wall (OFW) method are 86.764 KJ/mol for the activation energy ( $E_a$ ) and  $1.29 \times 10^{10} \text{ min}^{-1}$  for the pre exponential factor (A).

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# CHAPTER 1

## INTRODUCTION

### 1.1 BACKGROUND STUDY

One of the most contributors to municipal solid waste is plastic (Das & Tiwari, 2017). The most common type of plastic is Polyethylene terephthalate or commonly abbreviated as PET or PETE. For the past year, there is a drastic growth of plastic industry especially in the production Polyethylene terephthalate and polystyrene. In 1990, the production plastic was estimated to be 80 million tonnes and the consumption of plastic material in Western Europe in year 2004 was 43.5 million tonnes (Al-Salem & Lettieri, 2010).

As the population in Malaysia increase, the plastic consumption also increased. Plastic and the other solid waste usually disposed by two method which are landfilling and incineration. However this two method is no more relevant due to the lack of available land for dumping (Das & Tiwari, 2017). PET waste plastic is combination of hydrocarbon, benzene and oxygen and PET plastic has high percentage of oxygen content.

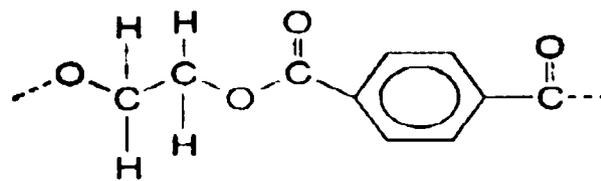


Figure 1.1 Chemical structure of PET



Figure 1.2 Symbol of PET