

**ANODIC ALUMINUM OXIDE-SUPPORTED GOLD
CATALYST (Au-AAO) FOR *p*-NITROPHENOL
REDUCTION**

NURUL HANIS NABILAH BINTI AHMAD FAIZAL

**BACHELOR OF SCIENCE (Hons.)
APPLIED CHEMISTRY
FACULTY OF APPLIED SCIENCES
UNIVERSITI TEKNOLOGI MARA**

AUGUST 2024



UNIVERSITI
TEKNOLOGI
MARA

Fakulti
Sains Gunaan

**SUBMISSION FOR EVALUATION
FINAL YEAR PROJECT 2 - RESEARCH PROJECT**

**ANODIC ALUMINUM OXIDE-SUPPORTED GOLD CATALYST (Au-AAO)
FOR
p-NITROPHENOL REDUCTION**

Name : Nurul Hanis Nabilah binti Ahmad Faizal
Student ID : 2021816912
Program : Bachelor of Science (Hons.) Applied Chemistry
Course code : FSG671
Mobile Phone :
E-mail : 2021816912@student.uitm.edu.my

** Please attach the Turnitin summary report, with your name clearly stated, at the end of your report and submit it together.*

Approval by Main Supervisor :

I certify that the work conducted by the above student is completed and approve this report to be submitted for evaluation.

Supervisor's name : Hanani binti Yazid
Date : 25 July 2024
Turnitin Similarity % : 3
Signature :

ANODIC ALUMINUM OXIDE-SUPPORTED GOLD CATALYST (Au-AAO) FOR *p*-NITROPHENOL REDUCTION

NURUL HANIS NABILAH BINTI AHMAD FAIZAL

**Final Year Project Report Submitted in
Partial Fulfilment of the Requirements for the
Degree of Bachelor of Science (Hons.) Applied Chemistry
in the Faculty of Applied Sciences
Universiti Teknologi MARA**

AUGUST 2024

TABLE OF CONTENT

ACKNOWLEDGEMENT	iii
TABLE OF CONTENT	iv
LIST OF TABLES	vii
LIST OF FIGURES	viii
LIST OF PLATES	x
LIST OF SCHEMES	xi
LIST OF ABBREVIATIONS	xii
LIST OF SYMBOLS	xv
ABSTRACT	xvi
ABSTRAK	xvii
CHAPTER 1 INTRODUCTION	
1.1 Background of Study	1
1.2 Problem Statement	2
1.3 Research Questions	3
1.4 Objectives of Study	4
1.5 Significance of Study	4
1.6 Scope and Limitation	5
CHAPTER 2 LITERATURE REVIEW	
2.1 Anodic Aluminum Oxide (AAO)	6
2.1.1 Fabrication of Anodic Aluminum Oxide (AAO)	8
2.1.2 Structural Form of Anodic Aluminum Oxide (AAO)	9
2.1.2.1 AAO Nanoporous Membrane	9
2.1.2.2 AAO Wires	11
2.2 Metal Nanoparticles: Supported Gold (Au) Catalyst	13
2.2.1 Synthesize Method of Supported Gold (Au) Catalyst	13
2.2.1.1 Co-Precipitation (CP)	14
2.2.1.2 Impregnation (IMP)	15
2.2.1.3 Deposition-Precipitation (DP)	17
2.2.2 Properties of Gold (Au) Catalyst	19
2.2.2.1 Size-Dependent Optical	20

ABSTRACT

ANODIC ALUMINUM OXIDE-SUPPORTED GOLD CATALYST (Au-AAO) FOR *p*-NITROPHENOL REDUCTION

Anodic aluminum oxide (AAO) demonstrates significant potential as catalyst support due to its strong mechanical properties, excellent thermal stability and adjustable pore size. However, the efficiency of AAO in wire form under similar anodization conditions remains unexplored. This study aims to evaluate the efficiency of AAO in nanoporous membranes and wire forms as a support for gold (Au) for the catalytic reduction of *p*-nitrophenol (*p*-NP). Both AAO forms were fabricated using an electrochemical anodization method with identical anodization parameters (electrolyte, voltage, and temperature) followed by a deposition-precipitation (DP) technique to deposit gold onto the AAO surface producing Au/mAAO and Au/wAAO catalysts. The Au/AAO catalysts were characterized using Fourier Transform Infrared Spectroscopy (FTIR) revealing shifts from 3453 cm^{-1} to 3461 cm^{-1} for Au/mAAO and 3434 cm^{-1} to 3438 cm^{-1} for Au/wAAO correspond to O-H group due to gold attachment. The catalytic activity of Au/mAAO and Au/wAAO were assessed by Ultraviolet-Visible Spectroscopy (UV-Vis), which showed rate constants (k) of $4.24 \times 10^{-3} \text{ s}^{-1}$ for Au/mAAO and $4.31 \times 10^{-4} \text{ s}^{-1}$ for Au/wAAO. According to the result, Au/mAAO shows better activity for the catalytic reduction of *p*-nitrophenol. It was subsequently analyzed using Field-Emission Scanning Electron Microscopy (FE-SEM). FE-SEM image revealed that Au/mAAO has closely packed and regular-shaped holes with an average AAO pore size of $76.75 \pm 11.29 \text{ nm}$ and size of Au nanoparticles (NPs) scattered near the pore region had an average size of $16.03 \pm 5.54 \text{ nm}$. Hence, future research should focus on developing more efficient catalysts by exploring other AAO structures and utilizing advanced laboratory instruments such as X-Ray Diffraction (XRD) and Inductive Coupled Plasma Optical Emission Spectroscopy (ICP-OES) for characterization.