SYNTHESIS, CHARACTERIZATION AND ANTIBACTERIAL ACTIVITY OF Co(II), Cu(II) AND Cd(II) TRANSITION METAL COMPLEXES OF NOS TRIDENTATE SCHIFF BASE DERIVED FROM S-2-FLUOROBENZYLDITHIOCARBAZATE AND ISATIN

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ABSTRACT

SYNTHESIS, CHARACTERIZATION AND ANTIBACTERIAL ACTIVITY OF Co(II), Cu(II) AND Cd(II) TRANSITION METAL COMPLEXES OF NOS TRIDENTATE SCHIFF BASE DERIVED FROM S-2-FLUOROBENZYLDITHIOCARBAZATE AND ISATIN

A new Schiff base was formed from the condensation reaction of S-2-fluorobenzyldithiocarbazate with isatin. Complexes of cobalt(II), copper(II) and cadmium(II) with Schiff base were successfully prepared. These compounds were characterized by elemental analysis and various physico-chemical techniques. The analysis of IR data shows that the Schiff base was coordinated to the copper(II) and cadmium(II) ions as NS bidentate ligands whereas in cobalt(II) ion as NOS tridentate ligand. The Schiff base and its metal complexes were screened for antibacterial activity by using disc diffusion method. Antibacterial activity screening was carried out against Gram positive and Gram negative bacteria which are *Bacillus subtilis* and *Pseudomonas aeruginosa*. The metal complexes were found to be strongly active against these selected bacteria compared to Schiff base.

CHAPTER 1

INTRODUCTION

1.1 Schiff base

Schiff base or azomethine is a compound that contains a carbon-nitrogen double bond with the nitrogen atom connected to an aryl or alkyl group (Imran *et al.*, 2007). Schiff bases are of the general formula $R_1R_2C=N-R_3$, where R_3 is an aryl or alkyl group that makes the Schiff base a stable imine. Schiff base also derived from an aniline, where R_3 is a phenyl or substituted phenyl known as anil. Schiff bases can be synthesized from an aromatic amine and a carbonyl compound by condensation reaction.

 R_1 - CR_2O + R_3 - NH_2 \rightarrow R_1 - CR_2 = N- R_3 + H_2O

Aldehyde Primary amine Schiff base water

 $\mathbf{R}_1, \mathbf{R}_2, \mathbf{R}_3 =$ aryl, alkyl or phenyl group

Figure 1.1 General formula of Schiff base (Parekh et al., 2005)