

UNIVERSITI TEKNOLOGI MARA

**DESULPHURISATION OF MAE MOH LOW-RANK
COAL USING ULTRASONIC WAVES AND
MICROWAVE IRRADIATION TREATMENT
PROCESSES WITH MIXTURE OF HYDROGEN
PEROXIDE AND HYDROCHLORIC ACID**

WAN IZHAN NAWAWI BIN WAN ISMAIL

**Thesis submitted in partial fulfillment of the requirements
for the degree of
Master of Science**

Faculty of Applied Sciences

May 2007

ABSTRACT

Mixture of 6% hydrogen peroxide and 0.1N hydrochloric acid (3:7 volume mixed ratio of peroxide to acid) was used to desulphurise Mae Moh low-rank high sulphur coal in the presence of three different processes i.e. ultrasonic waves, microwave irradiation and conventional magnetically stirring processes. The parameters that were used to investigate the effect of sulphur and ash removal from the coal using the three different processes were leaching time (i.e. 0.25, 0.5, 1, 2, 4 and 8 hour for both ultrasonic waves and magnetically stirring processes, and 0.5, 1, 2, 4, 10 and 20 minutes for microwave irradiation process), leaching temperature (30, 50 and 70°C; for both ultrasonic wave and magnetically stirring), and irradiation power (150, 700 and 100 watts for microwave irradiation). The success of the three different processes toward the total sulphur removal was measured by comparing the S/C atomic ratios of the treated coal with the untreated coal. The percent of inorganic sulphur removal in the treated coal was determined by using ASTM D2492, with organic sulphur being determined by the difference. In general, the treatments of coal sample using three different processes led to *ca.* 35-68, 29-93, 43-100, 2-50% and 37-71% of total sulphur, pyrite, sulphate, organic and ash removal, respectively. The optimal temperature and time for desulphurisation using ultrasonic waves were at 50°C and 0.5 hour, respectively, and for the magnetically stirring process were at 50°C and 0.5 - 8 hour, respectively. The optimal power and time for the microwave irradiation process, however, was at 700 watt and 4 minute, respectively. By comparing the three different processes, the ultrasonic waves treatment at 50°C for 0.5 hour was found to be the optimum condition for sulphur and ash removal with approximately 65, 82, 83, 48 and 55% of total sulphur, pyrite, sulphate, organic and ash reduction were achieved, respectively, without affecting the coal heating value. Moreover, pyrolysis on the treated coal samples via thermogravimetry analysis showed a slight increase in reactivity with comparison to the untreated coal sample. The by-product of coal desulphurisation contained humin compound and can be used in fertilizer industry. Further, the calorific values of treated coal samples were comparable to the untreated coal indicating no significant alteration in the coal macro-structure.

ACKNOWLEDGEMENT

IN THE NAME OF ALLAH, THE MOST GRACIOUS AND MOST MERCIFUL

Thanks to Allah S.W.T for giving me the opportunity to do and complete this thesis successfully. A journey is easier when you travel together. Interdependence is certainly more valuable than independence. This thesis is the results of two years of work whereby I have been accompanied and supported by many people. It is a pleasant aspect that I have now the opportunity to express my gratitude for all of them.

The first person I would like to thank is my supervisor; Assoc. Prof. Dr. Hj. Khudzir Hj. Ismail whose help, stimulating suggestions and encouragement helped me in all time of research for and writing of this thesis. Also not to forget to Mr. Mohd Azlan Mohd Ishak who kept an eye on the progress of my work and always available when I needed his advises.

I would also like to extend my thanks to Ministry of Science Technology & the Innovation (MOSTI), Malaysia for funding my studied (PASCA scheme) and the research (IRPA Grant: 02-02-01-0012-EA0012). To Prof. Dr. Pattarapan Prasassarakich from University of Chulalongkorn, Thailand for providing the coal sample.

Special thanks also forwarded to all the group members ;Zubri Zakaria, Mohd Fauzi Abdullah, Ridzuan Mat, Rafidah Rakal and Nur Nasulhah Kasim for their kindness and generous assistant throughout this research and to also those had contributed either directly or indirectly to this research. Their co-operations are not easily forgotten.

Finally, I would like to express my special gratitude to my dearest family especially lovely parents; Wan Ismail Wan Hamat and , for my wife; Nur Nadina Anuar, my brothers and sisters; Wan Izani, Wan Izudin, Wan Norlida and Wan Noraizon for their love, understanding, never ending encouragement and financial support to proceed and complete my research and studies in Universiti Teknologi MARA.

TABLE OF CONTENTS

	Page
Abstract	i
Acknowledgments	ii
Table of contents	iii
List of Tables	vi
List of Figures	viii
List of Plates	xi
List of Appendices	xii
Glossary	xiii
1.0 Introduction	
1.1 The Potential Energy from Coal	1
1.2 Important of Desulphurisation Technique and Problems Statement	4
2.0 Literature Review	
2.1 Coal Formation and Classification	7
2.2 Origin of Sulphur and Mineral Matter Forms in Coal	10
2.3 Humic Substance in Coal	12
2.4 Coal Assays	
2.4.1 Proximate analysis	13
2.4.2 Ultimate analysis	14
2.4.3 Sulphur Distribution Analysis	14
2.5 Chemical Coal Desulphurisation Process	
2.5.1 Mild Oxidation Techniques	15
2.5.2 Mild Reduction Techniques	17
2.5.3 Chemical Desulphurisation using Strong Bases	18
2.5.4 Chemical Desulphurisation using Strong Organic Bases	20
2.6 Application of Ultrasonic Waves for Sulphur Removal in Coal	21

CHAPTER 1

INTRODUCTION

1.1 The Potential Energy from Coal

Coal is one of the world's most abundant fossil fuels and an important source of energy. Historically, China was the first country in the world to use coal. It has been used for three thousand years for small purposes like smelting metals. Coal became more important as an energy source in the early twentieth century when the coal industry was developed in western European countries for different burning purposes which included heating, cooking, harnessing mechanical power, powering steam trains and ships (transportation) and generating heat for the manufacturing process [1].

Petroleum oil was found in the mid twentieth century and there has been an overdependence on this source as a fuel. Hence the potential of coal has not been fully exploited and coal demand has decreased considerably [2]. However, the overdependence on just only one non-renewable source (petroleum oil) as energy has contributed to the current world energy crisis.

Malaysia's National Energy Policy (1979) aims to have an efficient, secure and environmentally sustainable supply of energy in the future as well as to have an efficient and clean utilisation of energy. This strategy was formulated due to the aftermath of the two international oil crisis quantum leaps in price in 1973 and 1979 [3].

Malaysia is committed to manage its energy resources, ranging from fossil fuels to renewable resources. In 1981, the Four-Fuel Diversification Policy was designed to reduce the country's overdependence on oil as an energy source. The policy focuses on four main sources of fuel, namely coal, oil, gas and hydro, aimed at ensuring their