

UNIVERSITI TEKNOLOGI MARA

**FABRICATION OF HUMIDITY
SENSOR USING TA-DOPED TiO₂
NANOFLOWER AND TA-DOPED
TiO₂ NANOFLOWER/RGO
NANOCOMPOSITE**

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ABSTRACT

In this work, the feasibility of using nanostructured titanium dioxide (TiO_2) film as a sensing element of a resistive-type humidity sensor was explored. Flower-like TiO_2 nanostructure have been successfully prepared on glass substrate using a combination of radio frequency (RF) magnetron sputtering and modified solution immersion methods, eliminating the need for fluorine-doped tin oxide (FTO) substrate and autoclave. The TiO_2 seed layer was deposited on a glass substrate using RF magnetron sputtering to replace the FTO substrate for TiO_2 nanostructures growth. The properties of the prepared flower-like TiO_2 nanostructure films were characterised using field emission scanning electron microscopy (FESEM), energy dispersive X-ray spectroscopy (EDS), high-resolution transmission electron microscopy (HR-TEM), X-ray photoelectron spectroscopy (XPS), X-ray diffraction spectroscopy (XRD), Raman spectroscopy, reverse double beam photoacoustic spectroscopy (RBD-PAS), Hall-effect measurement, contact angle measurement, and humidity sensor response measurement. Humidity sensor fabricated using this flower-like TiO_2 nanostructure has showed excellent sensor response due to the immense surface area provided by the unique structure. The effect of Tantalum (Ta) doping on the characteristics of TiO_2 nanoflower was also examined. Evolution of TiO_2 structure from nanoflower to microsphere was observed with the addition of Ta, leading to the improvement of humidity sensor response. The optimum Ta doping concentration is at 3.0 at.% which yielded humidity sensor response of 53,909%. Reduced graphene oxide (rGO) has also been added to the TiO_2 nanoflowers to produce nanocomposite with enhanced humidity sensing performance. A solution containing rGO was prepared and drop-casted onto TiO_2 nanoflowers. The nanocomposite recorded maximum sensor response of 39,590%. Finally, to assess the combined effect of Ta-doping and rGO addition, Ta-doped TiO_2 microsphere/rGO nanocomposite have been prepared. The humidity sensor fabricated using the material exhibited maximum humidity sensor response of 232,152%. In summary, the improved humidity sensor performance can be attributed to enhanced surface area, increased oxygen vacancy sites, reduced electrical resistivity, and formation of Schottky junction. The findings from this study suggest that this type of material has the potential to be employed to fabricate high sensitivity humidity sensing device.

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CHAPTER ONE

INTRODUCTION

1.1 Research Background

Humidity is one of the important parameters affecting our everyday life. It is defined as the measurement of the amount of water molecules in our surrounding. Accurate measurement of this parameter is becoming increasingly crucial as the monitoring of humidity is needed in various fields including food preservation, agriculture, semiconductor fabrication, weather forecasting, health screening and so on. Humidity is often measured using humidity sensor or hygrometer. There is an apparent need to improve the sensitivity of humidity sensor in order to cater for the industry demand.

1.2 Titanium Dioxide (TiO₂): Characteristics and Applications

Titanium dioxide (TiO₂) is a compound semiconductor with multitude of unique characteristics which makes it stands out as the popular material in various application. It is reported that in the year 2019 alone, the production of TiO₂ have surpasses 3 billion tons [1] with China as its main producer. It is an environmentally friendly material with high chemical and physical stability. It is also hydrophilic in nature due the presence of Ti³⁺ and oxygen vacancies defect sites. The bandgap energy is relatively wide, at around 3.0-3.2 eV [2].

This transition metal oxide is also known as titania and its molecules comprise of a Ti atom and two O atom. It can occurs in three phases namely anatase, rutile, brookite [3]. Both anatase and rutile have tetragonal structure while the brookite is of orthorhombic structure as shown in Figure 1.1. In both types, octahedron of 6 O²⁻ ions surround Ti⁴⁺ ion [4]. The stable nature of both anatase and rutile enable them to be frequently used in various applications.