

SIIC084

A COMPARATIVE STUDY ON THE EFFECT OF PALLADIUM DOPANT CONCENTRATION ON THE PHOTOCATALYTIC PROPERTIES OF TIN(IV) OXIDE

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Abstract:

Several type of semiconductor photocatalysts such as TiO₂, ZnO and SnO₂ have been applied in treating dye containing wastewater. However, the application of SnO₂ metal oxides as photocatalysts still remain limited despite the potential of the material to oxidize and degrade pollutant materials in wastewater. SnO₂ can be modified with transition metal dopant such as palladium to improve its photocatalytic properties especially in the visible region of solar spectrum. In this work, SnO₂ with different concentration of palladium dopant (0.5, 1, 3 and 5 mol%) were successfully analysed. The characterization data from X-ray diffraction (XRD), Fourier-Transform infrared spectroscopy (FTIR), high-resolution transmission electron microscopy with selected area electron diffraction (HRTEM-SAED), field emission scanning electron microscopy (FESEM) and UV-vis spectroscopy were carefully analysed and compared with previous studies. From XRD analysis, the results showed strong evidence of SnO₂ nanoparticles formation. Whereas the FTIR peaks further confirmed the presence of the Sn-O bonds. The UV-vis result revealed the band gap of Sn-O considerably decreased from 3.75 eV to 3.66 eV with increasing Pd concentration. The selected 5 mol% Pd doped SnO₂ was tested on photocatalyst reaction with methylene blue. In addition, the image of Pd doped SnO₂ from HRTEM-SAED demonstrated the SnO₂ nanoparticles existed in short nanorods in a closely packed structure. The comparison with literature review shows that Pd doping improve the photocatalytic properties of SnO₂.

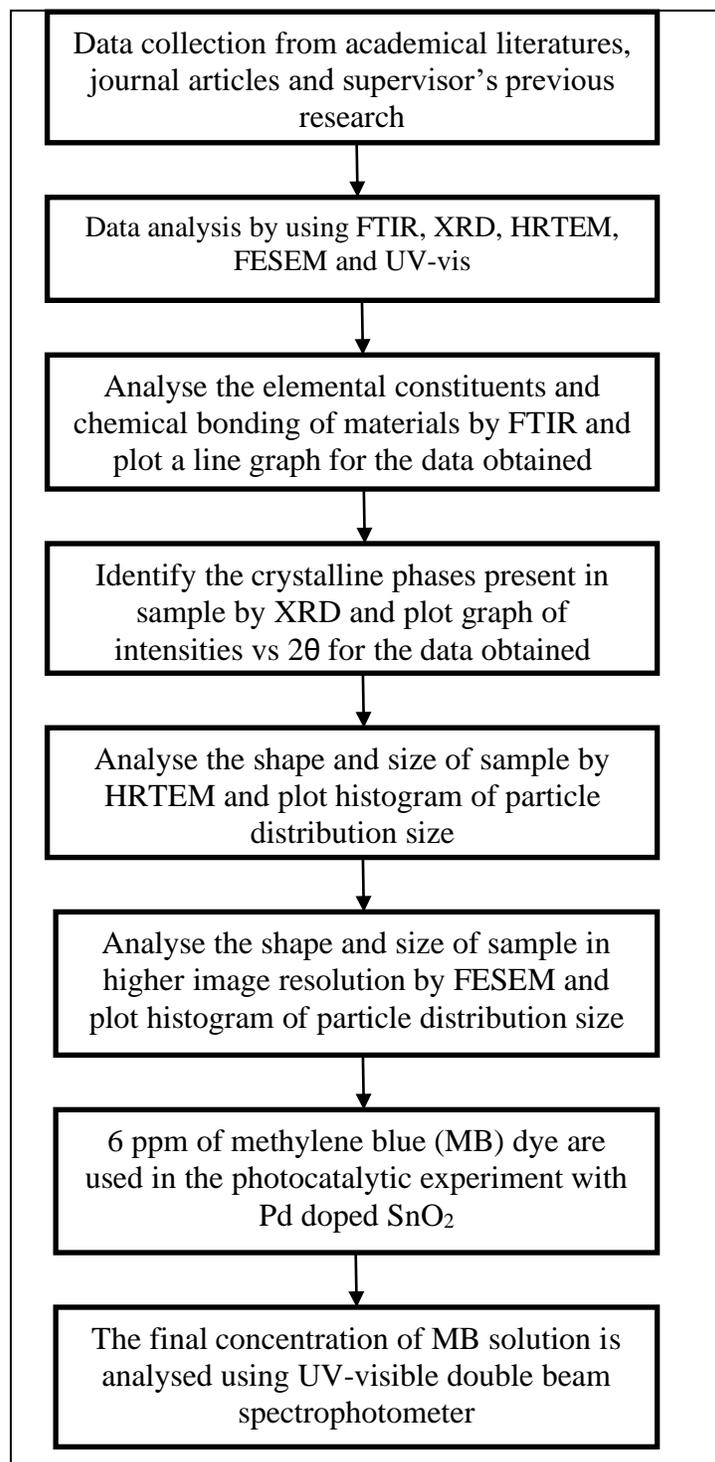
Keywords:

SnO₂, Pd doped, nanorods, photocatalyst, concentration

Objectives:

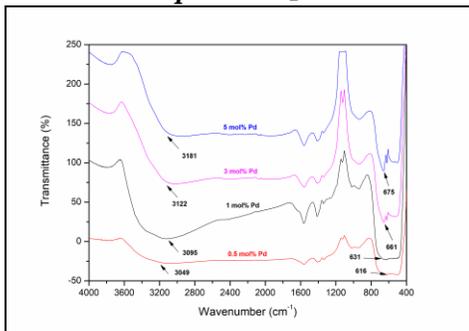
- To analyse the effect of Pd concentration on the chemical and structural properties of SnO₂.
- To compare the results of Pd-doped SnO₂ characterization with previous data (XRD, FTIR, HRTEM-SAED, FESEM and UV-Vis).
- To propose a plausible mechanism for doped SnO₂ photocatalyst based on the study.

Methodology:

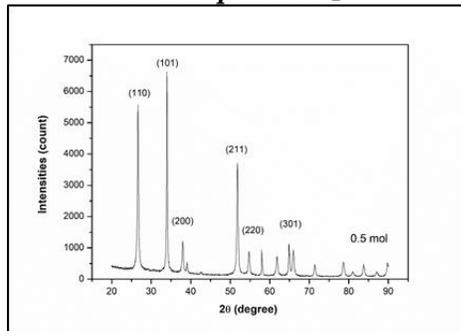


Results:

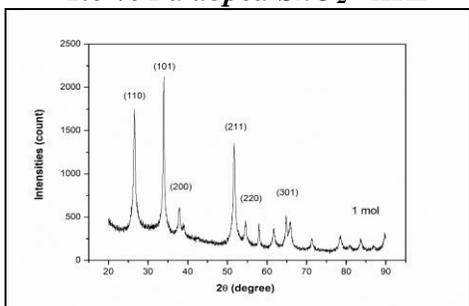
Pd doped SnO₂ - FTIR



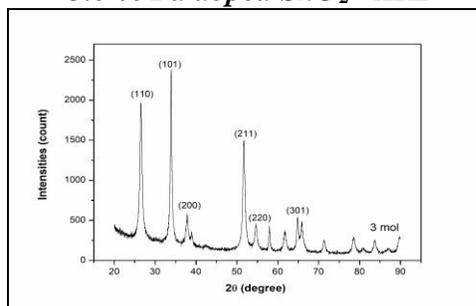
0.5 % Pd doped SnO₂ - XRD



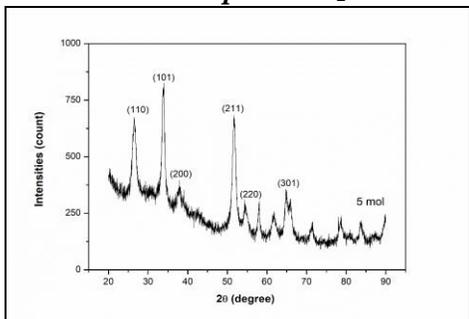
1.0 % Pd doped SnO₂ - XRD



3.0 % Pd doped SnO₂ - XRD



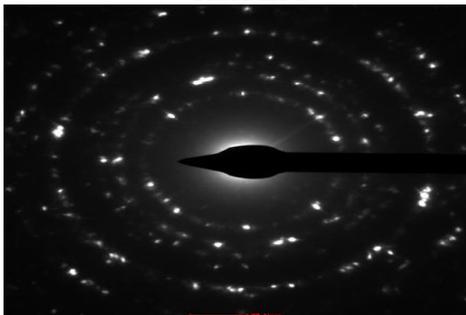
5.0 % Pd doped SnO₂ - XRD



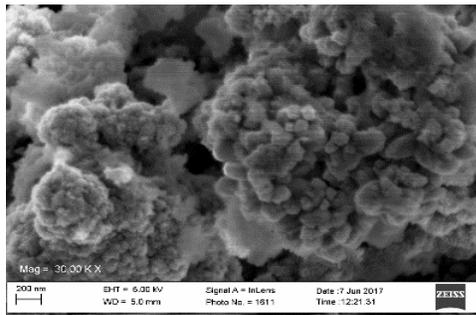
Pd doped SnO₂ - HRTEM (10 nm)



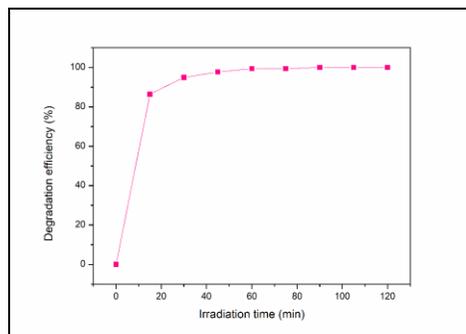
SAED pattern of Pd doped SnO₂ - HRTEM



Pd doped SnO₂ - FESEM (30.00 kx)



Photocatalytic experiment



Conclusion:

The properties of SnO₂ nanorods with different concentration of Pd dopant (0.5%, 1%, 3% and 5%) have successfully analysed in this study. The FTIR shown a significance results in the photocatalytic properties of SnO₂ material. From the FTIR results, the FTIR peaks produced were attributed mainly on the O-Sn-O, Sn-O and OH bonds. Whereas in XRD analysis, the results of XRD pattern showed strong evidence for the formation of SnO₂ nanoparticles. In addition the structure of SnO₂ nanoparticles were confirmed in tetragonal rutile structure with no other obvious characteristics impurity peaks observed from the XRD pattern. Moreover, intensity at major peaks (110,) (101) and (211) plane for the Pd doped SnO₂ seen to be reduce with higher concentration of Pd dopant. In this work it have been analysed that the Pd-SnO₂ has exhibited changes in the physical properties which produce an alterations on the crystal structure, morphology, energy band gap as well as the photocatalytic activity. The image of Pd doped SnO₂ from HRTEM-SAED demonstrated the as-prepared SnO₂ nanoparticles existed in polycrystalline nanorods structure with an average crystal size measured of 11.6 nm ± 1.8797 nm. In contrast, FESEM images analysed the samples are mostly short nanorods in closely packed structure and agglomerated with each other. The measured size of the particle is approximately in the range of 19.8 nm ± 2.3073 nm. It is clearly seen that doping SnO₂ with a moderate amount of Pd metal affects the shapes of SnO₂ slightly. The size of the particles is reduced upon increasing Pd concentration caused by Pd doping. On the other hand, the band gap of Sn-O is also reduced with increasing Pd concentration. The maximum degradation efficiency of methylene blue is 100% in 120 min. The narrowing of band gap of SnO₂ provide increase in absorption of visible light irradiation. The 5% concentration of Pd dopant suggested to be suitable concentration for the degradation of MB dye.