

UNIVERSITI TEKNOLOGI MARA

**SUGARCANE BAGASSE ASH
SUPPORTED CALCIUM OXIDE AS
THE HETEROGENEOUS BASIC
CATALYST FOR BIODIESEL
PRODUCTION**

AUNIE AFIFAH BINTI ABDUL MUTALIB

MSc

May 2021

AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the results of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

I, hereby, acknowledge that I have been supplied with the Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.

Name of Student : Aunie Afifah Binti Abdul Mutalib

Student I.D. No. : 2018832462

Programme : Master of Science (Chemistry) – AS756

Faculty : Applied Sciences

Thesis Title : Sugarcane Bagasse Ash Supported Calcium Oxide
as the Heterogeneous Basic Catalyst for Biodiesel
Production

Signature of Student :

Date : May 2021

ABSTRACT

Calcium oxide (CaO) is a high potential heterogeneous catalyst for biodiesel production. However, it is often reported with leaching problem requiring the usage of catalyst support which can further increase the catalyst production cost. Hence, in this study, the feasibility of the waste derived support which is the sugarcane bagasse ash (SCBA) impregnated with CaO as a heterogeneous basic catalyst in biodiesel production was explored. The SCBA was successfully prepared via calcination process for 2 hours under various temperatures at 500 °C, 600 °C, 700 °C and 800 °C, and were impregnated with varied CaO loading ranging from 10 wt.% to 40 wt.%. The prepared CaO/SCBA catalysts were furthered qualitatively and quantitatively characterized using Fourier Transform-Infrared Spectroscopy (FTIR), Scanning Electron Microscopy (SEM), X-Ray Diffraction (XRD), Temperature Programmed Desorption of Carbon Dioxide (TPD-CO₂), Thermal Gravimetric Analysis (TGA), X-Ray Fluorescence (XRF) and Brunauer-Emmett-Teller (BET) surface techniques. The BET and TPD-CO₂ characterizations revealed that varied calcination temperature and CaO loading have significant impacts on the surface area and basicity of the catalyst. It was found that the increment of calcination temperature and CaO loading had decrease the surface area of the catalyst. Besides that, among the catalysts prepared, CaO_(40%)SCBA_{600°C} catalyst possessed the highest basic densities (3889.75 μmol g⁻¹) which further contributed to the highest biodiesel yield in the preliminary transesterification test. The CaO_(40%)SCBA_{600°C} catalyst was then further implemented into the optimization reaction condition steps which include various parameters (methanol-to-oil molar ratio, catalyst loading, reaction time and reaction temperature). It was found that, the catalyst has a high catalytic activity by producing 93.8% FAME yield at methanol-to-oil molar ratio of 20:1, reaction temperature of 65 °C, 6 wt.% of catalyst in 3 hours of reaction time. Furthermore, the reusability analysis found that the catalyst can be re-used up to 5 reaction cycles with biodiesel yield of from 93% at first cycle and drop to 70.3% on the fifth cycle. This may be contributed by the interaction between the impregnated CaO and silica in the SCBA forming CaSiO₄ as recorded by the previous XRD and FTIR analysis. This proved the significant role of the natural SiO₂ in the SCBA which helps to enhance the catalytic performance and also reduced the catalyst's deactivation problem by minimizing the leaching of active sites, CaO.

ACKNOWLEDGEMENT

Firstly, I wish to thank God for giving me the opportunity to pursue my master degree and experienced the journey. His blessings in terms of the emotional and physical strength have allowed me to complete this long and challenging journey successfully.

The infinite gratitude goes to my supervisor Dr. Mohd Lokman Bin Ibrahim for guiding me with sincerity, thoughtfulness, tolerance, patience, encouragement and knowledge. Thanks also to all my co-supervisors, Dr. Juan Bin Matmin and Professor Dr. Yun Hin Taufiq-Yap for all the supports provided.

I am also thankful with the aid from the all laboratory assistances and facilities that involved. Thus, my appreciation also belongs to Faculty of Applied Sciences and Institute of Sciences of University Teknologi MARA (UiTM) and Catalyst Science and Technology Research Centre, Faculty of Applied Sciences of University Putra Malaysia (UPM) for the accommodating supports and conveniences.

This journey is also fruitful and unforgettable with the presences of my dear friends and lab mates. Therefore, I would like to express a heartfelt credit to them for all the meaningful advice and moral encouragement.

Last but not least, this thesis is dedicated to my family especially my beloved parents for raising me and giving me a wonderful life.

TABLE OF CONTENTS

	Page
CONFIRMATION BY PANEL OF EXAMINERS	ii
AUTHOR'S DECLARATION	iii
ABSTRACT	iv
ACKNOWLEDGEMENT	v
TABLE OF CONTENTS	vi
LIST OF TABLES	x
LIST OF FIGURES	xi
LIST OF ABBREVIATIONS	xiv
CHAPTER ONE INTRODUCTION	1
1.1 Research Background	1
1.2 Problem Statement	2
1.3 Objectives	3
1.4 Significance of Study	3
1.5 Scope of Research	4
CHAPTER TWO LITERATURE REVIEW	5
2.1 Biodiesel as an Alternatives Fuel	5
2.1.1 The Generations of Biodiesel and Their Feedstocks	5
2.1.2 Biodiesel and its Advantages	7
2.2 Biodiesel Production Techniques	9
2.2.1 Transesterification Reaction	9
2.2.2 Other Biodiesel Conversion Methods	12
2.3 Catalyst in Biodiesel Production	14