SILVER OXIDE (AgO) deNO_x CATALYST IN THE STATIONARY FLUE GAS: FTIR AND TPR CHARACTERIZATION

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DECLARATION

"I hereby declare that this report is the result of my own work except for quotations and summaries which have been duly acknowledged."

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ABSTRACT

The purposes of this study are to prepare the $deNO_x$ catalyst (AgO/Al₂O₃-SiO₂) and to characterize the deNO_x catalyst using temperature programmed reduction (TPR) and fourier transform infrared spectroscopy (FTIR). The optimum reduction temperature and the functional group of deNO_x catalyst is identified. Besides that, the applying of low cost catalyst for this study which is silver oxide (AgO) can replace the present catalyst metal such as Platinum Group Metal (PGM) which is very expensive. The problem of NO_x emission from the stationary sources such as transportation, fuel combustion and industrial process is the main focus to develop the deNO_x catalyst. The methodology used to prepare catalyst includes ball-milled and pre- calcinations process and mechanical mixing catalyst. For calcinations process, five calcinations temperature involved which is 400°C, 500°C, 600°C, 700°C and 800°C respectively. Result for the TPR characterization show that the maximum reduction temperature occurs in the temperature range of 270-690° where the most gas hydrogen was absorbed. The total amounts of gas absorbed during this temperature range are 308.63 µmol/g. While for the FTIR the functional group that appear in the prepared deNO_x catalyst appear is Si-O bond because the ratio of wt% of the silica is higher than AgO and alumina.

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CHAPTER 1 INTRODUCTION

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