# PVC + (NH<sub>4</sub>)HSO<sub>4</sub> + SiO<sub>2</sub> (10 $\mu$ m) POLYMER ELECTROLYTES AND ITS ELECTRICAL PROPERTIES

### By:

### **NURHASHIDAH BINTI ANUAR**

## Supervisor:

### PROF. DR. RI HANUM YAHAYA SUBBAN

Co – supervisor:

DR. NOR SABIRIN MOHAMED

# FINAL YEAR PROJECT THESIS BACHELOR OF SCIENCES (Hons.) INDUSTRIAL PHYSICS FACULTY OF APPLIED SCIENCES UNIVERSITI TEKNOLOGI MARA MALAYSIA

**MAY 2011** 

This Final Year Project Report entitled 'PVC + (NH<sub>4</sub>)HSO<sub>4</sub> + SiO<sub>2</sub> (10 $\mu$ m) POLYMER ELECTROLYTES AND ITS ELECTRICAL PROPERTIES' was submitted by Nurhashidah Binti Anuar, in partial fulfilment of the requirements for the Degree of Bachelor of Science (Hons.) Industrial Physics, in the Faculty of Applied Sciences and was approved by

Prof Dr Ri Hanum Yahaya Subban Supervisor B. Sc. (Hons.) Industrial Physics Faculty of Applied Sciences Universiti Teknologi MARA 40450 Shah Alam Selangor

Dr. Nor Sabirin Mohamed
Co-Supervisor
Centre For foundation Studies in Science
Universiti Malaya Kuala Lumpur

Assc. Prof Dr. Md Yusof Theeran
Project Coordinator
B. Sc. (Hons.) Physics
Faculty of Applied Sciences
Universiti Teknologi MARA
40450 Shah Alam
Selangor

Tn. Hj. Isa Bin Mohd Yusof Head of Programme B. Sc. (Hons.) Industrial Physics Faculty of Applied Sciences Universiti Teknologi MARA 40450 Shah Alam Selangor

Date: 1 9 MAY 2011

### **ACKNOWLEDGEMENT**

Alhamdulillah, I am highly grateful to Allah S.W.T, for the strength given to me to complete my final year project thesis entitled 'PVC + (NH<sub>4</sub>)HSO<sub>4</sub> + SiO<sub>2</sub> (10μm) POLYMER ELECTROLYTES AND ITS ELECTRICAL PROPERTIES' which has been successfully submitted as planned. Firstly, I would like to express my thanks to my supervisor, Prof Dr. Ri Hanum Yahaya Subban for the continuous guidance, comments and advice in the course of this work. I would also like to express appreciation and thanks especially to Noor Shuhada, Nur Farhanan, Nurul Asmida and Siti Nor wardah for their good cooperation and support during this project. And not forgetting, special appreciation is dedicated to my family that gives me continuous support and encouragement.

# **TABLE OF CONTENTS**

CONTENTS  ACKNOWLEDGEMENT  TABLE OF CONTENTS  ABSTRACT  LIST OF TABLE		<u>PAGE</u> i ii- iii iv v			
			LIST OF	FIGURES	vi
			<u>CHAPTE</u>	<u>IR 1</u>	e u
			1.0	Introduction	1-2
			1.1	Problem Statement	2
1.2	Objectives	2			
1.3	Scope of Study	3			
1.4	Aims of Study	3.			
<u>CHAPTE</u>	<u>R 2</u>				
2.0	Introduction to Polymer Electrolyte	4-5			
2.1	Classification of Polymer Electrolyte				
	2.0.1 Dry Solid Polymer Electrolyte	5-6			
	2.0.2 Gel Polymer Electrolyte	6-7			
	2.0.3 Composite Polymer Electrolyte	7-8			
2.2	Polyvinyl Chloride (PVC) based electrolytes				

### **ABSTRACT**

Polymer electrolytes composed of PVC (polyvinyl chloride) as a host polymer, (NH<sub>4</sub>)HSO<sub>4</sub> as a salt and SiO<sub>2</sub>(10 $\mu$ m) as a filler will be prepared by solution cast technique using tetrahydrofuran (THF) as a solvent. The conductivity measurements of the samples were carried out using HIOKI-LCR Hi-Tester and are analyzed using impedance spectroscopy (IS). The sample with highest conductivity was recorded for 60 wt % PVC – 40 wt % (NH<sub>4</sub>)HSO<sub>4</sub> with value of 1.32E-08 Scm<sup>-1</sup>. The effects of addition of micro sized SiO<sub>2</sub> adder on the conductivity of the highest conducting any polymer electrolyte was investigated. The highest conductivity was obtained for 92 wt % PVC – (NH<sub>4</sub>)HSO<sub>4</sub> – 8 wt % SiO<sub>2</sub> with value of 3.23E-05 Scm<sup>-1</sup> which is an increase of 3.23E-05 orders of magnitude compared to the filler-free sample.