### THE EFFECT OF PH ON THE PROPERTIES OF SILICA NANOPARTICLES

HALAWATI BINTI OTHMAN

Final Year Project Report Submitted in Partial Fulfilment of the Requirements for the Degree of Bachelor of Science (Hons.) Physics in the Faculty of Applied Sciences Universiti Teknologi MARA

JANUARY 2020

# **TABLE OF CONTENTS**

			Page		
TAB LIST LIST LIST ABS	LE OF ( OF TA OF FIC	-	iii iv vi vii ix x xi		
СНА	PTER 1	INTRODUCTION			
1.1		round of the study	1		
1.2	-	m statement	3		
1.3	Signifi	icant of study	4		
1.4	Object	tives of study	5		
СНА	PTER 2	LITERATURE REVIEW			
2.1	Effect	of pH.	6		
2.2	Mesop	borous Silica.	7		
СНА	PTER 3	METHODOLOGY			
3.1	Introd	uction.	12		
3.2	Materi	ials	12		
3.3	Appar		13		
3.4		dology	13		
3.5		cterization Technique	16		
	3.5.1		16		
	3.5.2		17		
		Fourier Transform Infrared Spectroscopy (FTIR)	18		
		Ultraviolet-Visible absorption (UV-Vis)	19		
	3.5.5	X-ray Powder Diffraction (XRD)	21		
СНА		RESULTS AND DISCUSSION			
4.1	•	Energy Dispersive X-Ray Spectroscopy (EDX) 22			
4.2	Size and Morphology Analysis				
4.3	Fourier Transform Infrared Spectrometer (FTIR) analysis 28				

4.5 Structure Analysis from X-Ray Diffraction (XRD)	34		
CHAPTER 5 CONCLUSION AND RECOMMENDATIONS			
5.1 Summary	36		
5.2 Future research	37		
CITED REFERENCES			
APPENDIX			
CURRICULUM VITAE			

## LIST OF TABLES

Table	Caption	Page
3.1	Detailed parameter used for the reaction. NaOH changed from 0.10 mL to 0.30 mL.	15
4.1	Value of pH from different amount of NaOH	22
4.2	Percentage of element detected using EDX for uncalcined sample	25
4.3	Percentage of element using EDX for calcined sample	25
4.4	Chemical structure related to the functional group	30
4.5	Details of peak for calcined sample S0.10 and S0.20	35

#### ABSTRACT

#### THE EFFECT OF PH ON THE PROPERTIES OF SILICA PARTICLES

In this project, silica particles have been successfully produced by using adjusted Stöber method. Tetraethyl orthosilicate (TEOS) was used as source of silica, Sodium Hydroxide (NaOH) as catalyst, and Cetyl Trimethylammonium Bromide (CTAB) as pore-forming agent and surfactants. In order to study the effect of pH on the properties of silica particles, the amount of NaOH has been adjusted to find the different value of pH. The samples were analysed by using EDX, SEM, FTIR, UV-Vis, and XRD. According to SEM, the size of silica particles produced is about 0.423 micrometer. From UV-Vis spectroscopy, the direct energy band gap has been analysed by using Tauc plot. The energy band gap has been estimated for S0.10, S0.15, S0.20, S0.25 and S0.30 samples are 3.9184, 3.9554, 3.8679, 3.9939 and 4.0695 eV respectively. Increase of energy bandgap shows the decrease of particles size. XRD pattern shows several sharp peaks for sample S0.1 and S0.2 that defined as crystalline structure. While, for S0.30 sample show broad peak which determine amorphous pattern state of silica. The reaction parameters should be carefully adjusted in order to produce high quality silica particles which suitable for application in drug delivery.