

UNIVERSITI TEKNOLOGI MARA

**EFFECTS OF K⁺ AND Ni²⁺
SUBSTITUTION ON STRUCTURAL,
MAGNETIC, ELECTRICAL
TRANSPORT AND ELECTRONIC
PROPERTIES OF CHARGE
ORDERED MONOVALENT-DOPED
Pr_{0.75}Na_{0.25}MnO₃ MANGANITES**

ROZILAH BINTI RAJMI

Thesis submitted in fulfillment
of the requirements for the degree of
Doctor of Philosophy in Science

Faculty of Applied Sciences

December 2019

AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the results of my own work, unless otherwise indicated or acknowledged as referenced work. This thesis has not been submitted to any other academic institution or non-academic institution for any degree or qualification.

I, hereby, acknowledge that I have been supplied with the Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.

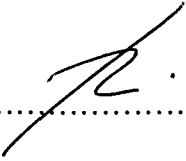
Name of Student : Rozilah Binti Rajmi

Student I.D. No. : 2014619282

Programme : Doctor of Philosophy (Science)– AS950

Faculty : Applied Sciences

Thesis Title : Effects of K^+ and Ni^{2+} Substitution on Structural, Magnetic, Electrical Transport and Electronic Properties of Charge Ordered Monovalent-Doped $Pr_{0.75}Na_{0.25}MnO_3$ Manganites.

Signature of Student : 

Date : December 2019

ABSTRACT

This study involves investigation on physical properties of charge ordered (CO) monovalent-doped orthorhombic $\text{Pr}_{0.75}\text{Na}_{0.25-x}\text{K}_x\text{MnO}_3$ ($x=0-0.20$) and $\text{Pr}_{0.75}\text{Na}_{0.25}\text{Mn}_{1-x}\text{Ni}_x\text{O}_3$ ($x=0-0.10$) manganites and density functional theory (DFT) computations of electronic properties of insulating $\text{Pr}_{0.75}\text{Na}_{0.25}\text{Mn}_{1-x}\text{Ni}_x\text{O}_3$ ($x = 0, 0.25$ and 0.50) manganites using LDA+ U method. The CO manganites were substituted with K^+ and Ni^{2+} at Na- and Mn-sites of the $\text{Pr}_{0.75}\text{Na}_{0.25}\text{MnO}_3$ respectively, to activate its electronic and magnetic properties. Both substitutions successfully suppressed the CO state and induced ferromagnetic to paramagnetic (FM-PM) transition, accompanied by metal to insulator (M-I) transition. The inducement of FMM-PMI transition for K^+ -substitution increased Curie temperature (T_C) and M-I transition temperature (T_{MI}) which was attributed to the reduction in MnO_6 octahedral distortion which leads to the enhancement of double exchange (DE) interaction. Meanwhile, for Ni-substituted samples the induced FMM-PMI transition caused lowering of T_C and T_{MI} which were suggested due to enhanced antiferromagnetic superexchange interactions involving $\text{Mn}^{3+}-\text{O}-\text{Mn}^{3+}$, $\text{Mn}^{4+}-\text{O}-\text{Mn}^{4+}$ and $\text{Ni}^{2+}-\text{O}-\text{Ni}^{2+}$ which decreased the ferromagnetic superexchange interaction between $\text{Ni}^{2+}-\text{O}-\text{Mn}^{4+}$ leading to reduction of DE interaction. DFT calculation of orthorhombic $\text{Pr}_{0.75}\text{Na}_{0.25}\text{MnO}_3$ with FM phase via LDA+ U with $U_{Mn}=2$ eV showed better agreement with the experimental crystal volume and bandgap results. Meanwhile, DFT calculations on orthorhombic $\text{Pr}_{0.75}\text{Na}_{0.25}\text{Mn}_{1-x}\text{Ni}_x\text{O}_3$ ($x = 0, 0.25$ and 0.50) with AFM phase with U value for Mn $3d$ set at 2 eV ($x = 0$) and 6 eV ($x = 0.25, 0.50$) while for both Pr $4f$ and Ni $3d$ the U values were set at 6 eV to compensate for the strong Coulomb repulsion of $3d$ electrons for Ni. The calculated crystal volume structure and bandgap for Ni free and Ni-substituted ($x = 0.25$ and 0.50) samples showed better agreement with the experimental data for the selected U values. DOS calculations revealed that all the FM and AFM samples showed half-metallic (HM) character. Partial DOS results showed Mn, Ni and O atoms contributed more significantly to the electronic states at E_F for the spin down channel with higher degree of hybridization between the Mn $3d$ /Ni $3d$ and O $2p$ electrons.

ACKNOWLEDGEMENT



To begin with, I would like to express my highest gratitude to God Almighty, Allah S.W.T for His endless blessings.

First and foremost, I would like to express my gratitude to my supervisor, Prof. Dr. Ahmad Kamal Hayati Yahya for his invaluable support, guidance and advice. His suggestion, criticism and willingness to motivate me contributed greatly towards the completion of this project. I would like to also thank my co-supervisors, Dr. Norazila Ibrahim and Dr. Muhamad Kamil Yaakob for their help and guidance to the success of the project and hence the thesis. I would also like to thank, Dr. Rosdiyana, Dr. Zakiah, Puan Syafawati and Encik Fauzi for sharing their knowledge and ideas.

I would also like to thank Puan Nor'Azian Daud (Laboratory's officer), Encik Abul Khamis Ishak (X-Ray Diffraction), Puan Asarnida Haron (Scanning Electron Microscope), and the Faculty of Applied Science's laboratory assistants who had helped me in numerous ways in accomplishing this research.

My special gratitude also to my postgraduate friends, Dr. Shabani, Dr. Nurul Syahidah, Nurul Huda, Siti Maryam, Nor Asmira, Muliana, Siti Wahidah, Aizzuddin, Hazri, Hafiz, Haziq and the members of density functional theory (DFT) group who have given me the motivation and spiritual support throughout my research work.

The financial support provided by the Ministry of Higher Education through Young Lecturer Scheme scholarship and research grant are also gratefully acknowledged and appreciated here. Besides, I would also like to thank the management of the Faculty of Applied Sciences and Universiti Teknologi MARA (UiTM) for providing good study environment and research facilities to complete this project.

Finally, an honorable mention goes to my beloved family; my mother, Rasadah Binti Yahaya, my father, Rajmi Bin Kamih and also my siblings (Rozita, Mohd Rahmad and Muhd Zamri) who had given me their love, prayers, support and encouragement through my study. Without their support, it will be impossible for me to complete this study successfully. I will be forever grateful for their understanding and love, especially through the duration of the study.

Thank you.

TABLE OF CONTENT

	Page
CONFIRMATION BY PANEL OF EXAMINERS	ii
AUTHOR'S DECLARATION	iii
ABSTRACT	iv
ACKNOWLEDGEMENT	v
TABLE OF CONTENT	vi
LIST OF TABLES	xi
LIST OF FIGURES	xiv
LIST OF ABBREVIATIONS	xxvi
CHAPTER: ONE INTRODUCTION	1
1.1 Research Background	1
1.2 Problem Statement of Study	4
1.3 Objective of Study	8
1.4 Significance of Study	9
CHAPTER: TWO LITERATURE REVIEW	10
2.1 Introduction	10
2.1.1 Crystal Structure	10
2.1.2 Crystal Field Splitting	11
2.1.3 Jahn-Teller (JT) Effect	12
2.2 Magnetic Exchange Interactions	14
2.2.1 Double Exchange (DE)	14
2.2.2 Super Exchange (SE)	15
2.3 Electrical Transport Properties in Manganites	16
2.3.1 Metallic Region	17
2.3.2 Insulating Region	18
2.4 Magnetic Properties in Manganites	20
2.5 Magnetoresistance (MR) in Manganites	22
2.6 Factors Affecting Physical Properties of Manganites	24