## SYNTHESIS AND CHARACTERIZATION OF GRAPHITE OXIDE AND 2-METHYL-4-CHLOROPHENOXY ACETIC ACID-GRAPHITE OXIDE (MCPA-GO) NANOCOMPOSITE WITH ITS CONTROLLED RELEASE PROPERTY

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## NORILYANI IZZATI BINTI HASANUDDIN

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Dr. Tn Sheikh Ahamad Izaddin Bin Sheikh Mohd Ghazali Supervisor B. Sc. (Hons.) Chemistry Faculty of Applied Sciences Universiti Teknologi MARA 72000 Kuala Pilah Negeri Sembilan

Nurul Huda binti Abdul Halim Project Coordinator B. Sc. (Hons.) Chemistry Faculty of Applied Sciences Universiti Teknologi MARA 72000 Kuala Pilah Negeri Sembilan

Mazni binti Musa Head of Programme B. Sc. (Hons.) Chemistry Faculty of Applied Sciences Universiti Teknologi MARA 72000 Kuala Pilah Negeri Sembilan

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#### ABSTRACT

## SYNTHESIS AND CHARACTERIZATION OF GRAPHITE OXIDE AND 2-METHYL-4-CHLOROPHENOXY ACETIC ACID – GRAPHITE OXIDE (MCPA-GO) NANOCOMPOSITE WITH ITS CONTROLLED RELEASE PROPERTY

The graphite oxide and 2-methyl-4chlorophenoxy acetic acid- graphite oxide (MCPA-GO) nanocomposite were successfully synthesized by using improved Hummer's method and ion-exchange method respectively. In this study, MCPA-GO nanocomposite was synthesized at various concentration of MCPA ranging from 0.1 to 0.7 M to lowering its toxicity and increase the drugs/herbicides loading efficiency. Meanwhile, the graphite oxide was prepared with the use the graphite powder as starting material. Graphite oxide and MCPA-GO nanocomposite were characterized using Fourier transform infrared (FTIR) spectroscopy, powder X-ray diffraction (PXRD) and Carbon Hydrogen Nitrogen and Sulphur analyzer (CHNS). The FTIR spectra of MCPA-GO nanocomposite was showed resemblance peaks of the MCPA and graphite oxide indicated the inclusion of MCPA into the graphite oxide. As for XRD pattern, there was increasing in the basal spacing of the nanocomposite from the graphite oxide which by 9.3 Å to 9.7 Å. Other than that, the percentage loading of MCPA in the nanocomposite was calculated to be 98.0 % (w/w) based on the CHNS result. Next, the controlled release of MCPA-GO nanocomposite was done in two different solution which were in sodium chloride solution (NaCl) and sodium carbonate solution (Na<sub>2</sub>CO<sub>3</sub>). The release of MCPA into these solution was found to be dependent to the anion in the order of  $Na_2CO_3 > NaCl$  with the percentage release of 66 % and 10 % respectively. It proved that this release property exhibits the potential application of graphite oxide as effective nanocarrier of herbicides.