SYNTHESIS AND CHARACTERIZATION OF MAGNETIC NANOPARTICLES (MNPs) COATED WITH POLYPYRROLE (PPy) FOR REMOVAL OF 2,4-DINITROPHENOL (2,4-DNP)

NOR AZEAN NADHIRAH BINTI MOHAMED SANI

Final Year Project Report Submitted in Partial Fulfilment of the Requirements for the Degree of Bachelor of Science (Hons.) Chemistry in the Faculty of Applied Sciences Universiti Teknologi MARA (UiTM)

JULY 2017

This Final Year Project Report entitled "Synthesis and Characterization of Magnetic Nanoparticles (MNPs) Coated with Polypyrrole (PPy) for the Removal of 2,4-Dinitrophenol (2,4-DNP) in" was submitted by Nor Azean Nadhirah Binti Mohamed Sani, in partial fulfilment of the requirements for the Degree of Bachelor Science (Hons.) Chemistry, in the Faculty of Applied Sciences, and was approved by

Jamil Monamed Sapari Surervisor

School of Chemistry and Environments Faculty of Applied Sciences Universiti Teknologi MARA 72000 Kuala Pilah Negeri Sembilan

Dr. Kavirajaa Pandian a/l Sambasevam Co Supervisor

School of Chemistry and Environments
Faculty of Applied Sciences
Universiti Teknologi MARA
72000 Kuala Pilah
Negeri Sembilan

Nurul Huda Abd. Halim
Project Coordinator
School of Chemistry and Environments

Faculty of Applied Sciences
Universiti Teknologi MARA

72000 Kuala Pilah Negeri Sembilan Mazni Musa
Head of Programme

Head of Programme

School of Chemistry and Environments

Faculty of Applied Sciences Universiti Teknologi MARA

72000 Kuala Pilah

Negeri Sembilan

Date: 1/8/2017

TABLE OF CONTENTS

ACKNOWLEDGEMENT TABLE OF CONTENTS LIST OF TABLES LIST OF FIGURES LIST OF ABBREVIATIONS ABSTRACT ABSTRAK						
СНАР	TER 1 INT	RODUCTION				
1.1	Background of Study					
1.2	Problem Statement					
1.3	Significance of Study					
1.4	Objecti	Objectives of Study				
СНАР	TER 2 LIT	TERITURE REVIEW				
2.1	Phenol	ic Compound	7			
	2.1.1	2,4- Dinitrophenol	8			
2.2		rrole (PPy)	10 11			
2.3	Magnetic Nanoparticles (MNPs)					
	2.3.1	Olive oil coated MNPs	15			
	2.3.2	Silane coated MNPs	16			
	2.3.3	2 1	16			
	2.3.4	Polyaniline (PANI) coated MNPs	18			
	2.3.5	Polypyrrole (PPy) coated MNPs	19			
СНАР	TER 3 ME	THODOLOGY				
3.1	Materia		21 21			
3.2	Charac	acterization Technique				
3.3	•	Synthesis of magnetic nanoparticle coated with Polypyrrole				
	(MNPs-PPy)					
3.4	Removal and Determination of Phenolic Compound					
	3.4.1	Effect of pH	24			
	3.4.2	Effect of Temperature and concentration	24			
	3.4.3	Effect of Contact Time	25			
	3.4.4	Effect of Amount of Adsorbent	25			
	3.4.5	Effect of Interfering Ions	26			
	3.4.6	Reusability	26			

CHAPT	TER 4 RE	SULTS AND DISCUSSION			
4.1	Charac	Characterization Technique			
	4.1.1	Fourier Transform Infrared Spectroscopy (FTIR)	27		
	4.1.2	X-Ray Diffraction (XRD)	30		
	4.1.3	Scanning Electron Microscope (SEM)	31		
4.2	Scanni	Scanning Analyte			
4.3	Effect	Effect of Parameter			
	4.3.1	Effect of pH	36		
	4.3.2	Effect of temperature and concentration	37		
	4.3.3	Effect of contact time	39		
	4.3.4	Effect of amount of adsorbent	41		
	4.3.5	Effect of Interfering Ions	42		
	4.3.6	Reusability	43		
СНАРТ	ΓER 5 CO	NCLUSION AND RECOMMENDATION			
5.1	5.1 Conclusion				
5.2	Recom	mendation	46		
	REFERE	INCES	47		
	DICES		51		
CURRI	CULUM	VITAE	61		

ABSTRACT

SYNTHESIS AND CHARACTERIZATION OF MAGNETIC NANOPARTICLES (MNPs) COATED WITH POLYPYRROLE (PPy) FOR REMOVAL OF 2,4-DINITROPHENOL

Magnetic nanoparticles coated with Polypyrrole (MNPs-PPy) was successfully synthesized in this study. The characterization of MNPs-PPy was determined by using Fourier Transform Infrared Spectroscopy (FTIR) analysis, Scanning Electron Microscope (SEM) and X-ray Diffraction (XRD). MNP has been coated with PPy was shown by the results of FT-IR, SEM and XRD. The result of FT-IR has showed the peak of Fe-O at 580 cm⁻¹. SEM result has showed the morphology of the surface of MNPs-PPy which have globular structure. In addition, XRD result has showed the MNPs-PPy is in the crystalline form. The removal efficiency was determined by using Ultra violet Visible Spectroscopy (UV-Vis). The synthesized MNPs-PPy was successfully removed the 2,4-Dinitrophenol (2,4-DNP) under several conditions. The first step is to determine the level of performance of the MNPs-PPy for the removal of 2,4-DNP. When the MNPs-PPy performance can reach up to 85 % removal, the next step can be proceed. The next step is to find the best condition for the removal to occur. There are several parameters that need to be done such as the effect pH, temperature and concentration, contact time, amount of adsorbent, interfering ions and reusability. The result showed 98.67 % removal of analyte at pH 6, 318 K, 10 ppm analyte, 75 minutes and 50 mg of adsorbents.