SYNTHESIS AND CHARACTERIZATION OF SCHIFF BASE LIGANDS FOR CORROSION INHIBITORS OF MILD STEEL

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APPROVAL SHEET

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TABLE OF CONTENT

	Page
APPROVAL SHEET	II
ACKNOWLEDGEMENT	III
TABLE OF CONTENT	IV
LIST OF TABLES	VI
LIST OF FIGURES	VII
LIST OF ABBREVIATIONS	IX
LIST OF EQUATIONS	X
ABSTRACT	XI
ABSTRAK	XII

CHA	APTER 1 INTRODUCTION	1
1.1	Background Study	1
	1.1.1 Corrosion Inhibitors	1
	1.1.2 Hydrochloric (HCl) Solution as Acidic Medium	3
	1.1.3 Schiff Base Ligands	3
	1.1.4 Mild Steel	4
1.2	Problem Statement	5
1.3	Significance of the Study	6
1.4	Objective of the Study	6
	· ·	

CHAPTER 2 LITERATURE REVIEW 7 7 2.1 Mechanism of Corrosion Inhibitors 2.2 Acidic Medium 9 2.3 Schiff Base Ligands 10 Weight Loss Method 2.4 12 2.4.1 Effect of inhibitor concentration 12 2.4.2 Effect of immersion time 13 2.4.3 Effect of acid concentration 13 2.4.4 Effect of temperature 13

CHA	CHAPTER 3 METHODOLOGY	
3.1	Materials and Chemicals	14
3.2	Apparatus	14
3.3	Synthesis of Schiff Base Ligands	15
	3.3.1 Synthesis of 4-hydroxybenzylideneanaline (AB1)	15
	3.3.2 Synthesis of 4-[(4-methylbenzylidene)amino]phenol (AB2)	16
3.4	Characterization of AB1 and AB2	18
3.5	Corrosion Inhibitors Method	20
	3.5.1 Preparation of Solution	20
	3.5.2 Weight Loss Method	21
СНА	PTER 4 RESULT AND DISCUSSION	24
41	Characterization of Ligands	24
	4.1.1 Elemental Analysis	25
	4.1.2 Fourier-Transform Infared Data for AB1 and AB2	26
	4.1.3 ¹ H NMR Spectrum Data for AB1 and AB2	28
4.2	Weight Loss Method	30
СПА	DTED 5 CONCLUSION AND DECOMMENDATIONS	22
CHA	APTER 5 CONCLUSION AND RECOMMENDATIONS	33
5.1	Conclusion	33
5.2	Recommendation for Future Work	35
СІТІ	ED REFERENCES	36
		40
APP	ENDICES	40
CUR	RICULUM VITAE	48

ABSTRACT

SYNTHESIS AND CHARACTERIZATION OF SCHIFF BASE LIGANDS FOR CORROSION INHIBITORS OF MILD STEEL

This study focused on synthesis of Schiff base ligands as corrosion inhibitors of mild steel in hydrochloric acid solution. Two Schiff base ligands which is 4hydroxybenzylideneanaline (AB1) and 4-[(4-methylbenzylidene)amino]phenol (AB2) have been successfully synthesized through the condensation process with the percentage yield of 58% and 96%, respectively. Further characterized using melting point determination, CHNS Analyzer, FT-IR Spectrometer and NMR Spectrometer. Then the mild steel was immersed in 15 mL of 1 M HCL without and with three difference concentration of Schiff base ligands solution which are 0.1 M, 0.01 M and 0.001 M for weight loss method. The final weight of mild steel were measured and calculated after 40 hours immersion. The value of corrosion rate decreased with an increasing concentration of Schiff base ligands. The result revealed that, the decreasing corrosion rate of inhibitor concentration will increase the inhibition efficiency as shown by AB2 with the 83.2% followed by AB1 which gave 67.2% at inhibitor concentration of 0.1 M. It was clear that the AB2 have a better efficiency than AB1 due to the methyl group that act as electron donating group that can changing the electron density and activate the aromatic ring.