

UNIVERSITI TEKNOLOGI MARA

**ELASTIC, STRUCTURAL AND
OPTICAL STUDIES ON $x\text{Na}_2\text{O}$ -(35-
 x) V_2O_5 -65 TeO_2 , $y\text{WO}_3$ -(40- y) Ag_2O -60 TeO_2
AND $z\text{Nb}_2\text{O}_5$ -(20- z) BaO -80 TeO_2
TELLURITE BASED GLASS SYSTEMS**

MUHAMMAD UMAIR BIN MOHAMAD

Thesis submitted in fulfillment
of the requirements for the degree of
Doctor of Philosophy

Faculty of Applied Sciences

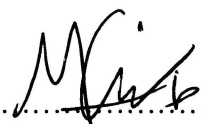
November 2015

AUTHOR'S DECLARATION

I declare that the work in this thesis was carried out in accordance with the regulations of Universiti Teknologi MARA. It is original and is the result of my own work, unless otherwise indicated or acknowledged as referenced work. This project has not been submitted to any other academic institution or non-academic institution for any other degree or qualification.

I, hereby, acknowledge that I have been supplied with the Academic Rules and Regulations for Post Graduate, Universiti Teknologi MARA, regulating the conduct of my study and research.

Name of Student : Muhammad Umair Bin Mohamad
Student ID No. : 2010327155
Programme : Doctor of Philosophy in Science
Faculty : Applied Sciences
Thesis Title : Elastic, Structural and Optical Studies of $x\text{Na}_2\text{O}$ -
($35-x$) V_2O_5 - 65TeO_2 , $y\text{WO}_3$ - $(40-y)\text{Ag}_2\text{O}$ - 60TeO_2
and $z\text{Nb}_2\text{O}_5$ - $(20-z)\text{BaO}$ - 80TeO_2 Tellurite Based
Glass Systems

Signature of Student : .....

Date : November 2015

ABSTRACT

Three glass systems with composition $x\text{Na}_2\text{O}-(35-x)\text{V}_2\text{O}_5-65\text{TeO}_2$ ($x=5, 10, 15, 20, 25$ mol %), $y\text{WO}_3-(40-y)\text{Ag}_2\text{O}-60\text{TeO}_2$ ($y=0, 10, 20, 30$ mol %) and $z\text{Nb}_2\text{O}_5-(20-z)\text{BaO}-80\text{TeO}_2$ ($z=0, 5, 10, 15$ mol %) were prepared by conventional solid state and melt-quenching methods. Elastic moduli and structural changes were studied by measuring ultrasonic shear and longitudinal velocities using the pulse-echo-overlap technique and Raman spectroscopy, respectively. For $x\text{Na}_2\text{O}-(35-x)\text{V}_2\text{O}_5-65\text{TeO}_2$ glass system, both longitudinal (v_L) and shear (v_s) velocities showed small steady decrease with addition of Na_2O from $x=5$ mol % to $x=15$ mol % followed by large decrease at $x>15$ mol %. Longitudinal modulus (C_L), shear modulus (μ), Young's modulus (Y), Hardness (H) and Debye Temperature (θ_D) also showed similar behavior to the ultrasonic velocities. The decrease in elastic moduli is suggested to be due to weakening of network rigidity of the glass system with increase in formation of non-bridging oxygen (NBO) as revealed by Raman spectroscopy. Additional analysis using bulk compression and ring deformation models showed that ratio between theoretical bulk modulus (K_{bc}) and experimental bulk modulus (K_e) was around 2.1 for $x=5-20$ mol % before an increase to around 2.4 for $x>20$ mol % indicating that the main compression mechanism was ideal isotropic compression. Meanwhile, optical band gap (E_{opt}) showed increase with Na_2O content and this is related to the increase of TeO_3 tp formation. For $y\text{WO}_3-(40-y)\text{Ag}_2\text{O}-60\text{TeO}_2$ glass system, the v_L and v_s showed large increase at $y=0-20$ mol % before dropping with further addition of WO_3 . Independent moduli (C_L and μ), K_e , Y and θ_D showed similar behaviors to both velocities. The large increase of the elastic moduli at $y=0-20$ mol % is suggested to be due to the increase in WO_6 octahedral indicating the increase of bridging oxygen (BO) and also formation of stronger Te-O-W bonds compared to Te-O-Te bonds. On the other hand, for $y>20$ mol %, the decrease in the elastic moduli was due to increase in NBO as a result of formation of WO_4 tetrahedral via breaking of Te-O-W network. Further analysis using bulk compression and ring deformation models showed a slight decrease in the ratio of ideal bulk modulus to experimental bulk modulus (K_{bc}/K_e) and average atomic ring size (l) for $y<20$ mol % followed by a large increase for $y>20$ mol %. Our analysis also indicates that limited ring deformation takes place and the main compression mechanism in this glass system was mainly ideal isotropic compression. On the other hand, optical band gap (E_{opt}) showed small variation for $y=0-20$ mol % but decreased upon higher WO_3 content while refractive index (n) showed the opposite trend. This optical behavior is suggested to be related to the changes in cross link density and NBO concentration in the glass system. For $z\text{Nb}_2\text{O}_5-(20-z)\text{BaO}-80\text{TeO}_2$ glass system, the v_L and v_s steadily increased with Nb_2O_5 content and are suggested to be influenced by independent moduli (C_L and μ). Elastic moduli such as K_e , Y and θ_D also showed similar behavior to the C_L and μ . The increase of elastic moduli is suggested to be due to the increase of bridging oxygen (BO) via TeO_4 tbp formation. On the other hand, Raman spectra showed increase in intensity of TeO_4 tbp and slight increase in NbO_6 octahedral.

Analysis on the spectra showed that the increase in BO is prominent compared to NBO. Optical energy gap (E_{opt}) was found to decrease with Nb_2O_5 which is suggested to be due to smaller difference between HOMO and LUMO states of TeO_4 tbp compared to that of TeO_3 tp and the averaging effect of E_{opt} of constituent oxides. Meanwhile, Urbach energy (E_u) decreased with Nb_2O_5 content indicating reduction in disorder of the glass structure.

TABLE OF CONTENTS

	Page
CONFIRMATION BY PANEL OF EXAMINERS	ii
AUTHOR'S DECLARATION	iii
ABSTRACT	iv
ACKNOWLEDGEMENTS	v
TABLE OF CONTENTS	vi
LIST OF TABLES	x
LIST OF FIGURES	xii
LIST OF SYMBOLS	xviii
CHAPTER ONE: INTRODUCTION	1
1.1 Research Background	1
1.2 Problem Statements	4
1.3 Objectives	6
1.4 Scope of Study	7
1.4 Significance of The Research	7
CHAPTER TWO: LITERATURE REVIEW	9
2.1 Introduction	9
2.2 Glass	9
2.2.1 Definations and Characteristics of Glass	9
2.2.2 Zachariasen's Rules	10
2.2.3 Classification of Glass	11
2.2.4 Advantageous and Application of Glass	12
2.3 Tellurite Glass	12
2.3.1 Structural Unit and Physical Properties of Tellurite Glass	14
2.3.2 X-ray Diffraction Pattern for Tellurite Glass	15