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27<sup>th</sup> APRIL 2024

UNIVERSITI TEKNOLOGI MARA  
CAWANGAN SELANGOR, KAMPUS DENGKIL  
MALAYSIA

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## Efficient Visible-Light-Active of $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$ photocatalyst for EDCs removal in wastewater

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### ABSTRACT

Photocatalysis has become more attractive and important since it has a great potential in solving environmental problems. One of the most important aspects of environmental photocatalysis is the selection of photo-active materials. Ag-based semiconductor materials are currently catching the interest and research efforts of numerous material researchers due to their wide range of applications especially photocatalyst for wastewater treatment. Silver carbonates ( $\text{Ag}_2\text{CO}_3$ ) is a common p-type semiconductor with a moderate band gap of 2.30 eV has also caught interest for its high-performance photocatalytic performance and anti-bacterial properties. To further improve the separation of photo-induced charge carriers,  $\text{Ag}_2\text{CO}_3$  was employed to couple with niobium pentoxide ( $\text{Nb}_2\text{O}_5$ ) can greatly accelerate the charge separation. In this work, hybrid  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite as a photo-active material was successfully prepared by a facile chemical precipitation method. The photocatalytic activities of the hybrid samples were evaluated by monitoring the photodegradation of bisphenol A (BPA) under visible light irradiation. The hybrid  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite sample exhibited promising results in the photocatalysis process with BPA removal of 79.09 % and showed good stability and reusability that can be used up to 4<sup>th</sup> cycle. The enhanced performance of hybrid  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  photocatalyst shows that it has potential for designing the dye waste-water treatment.

**Keywords:**  $\text{Ag}_2\text{CO}_3$ ;  $\text{Nb}_2\text{O}_5$ ; photocatalysis; visible; wastewater.

### 1. INTRODUCTION

A lot of emerging pollutants such as endocrine disrupting compounds (EDCs) are discharged into the aquatic system and posing a serious threat. One of the commonly known EDCs is Bisphenol A, an extremely harmful substance substantially produced in various plastic products (Godiya & Park, 2022). Hence, eliminating such hazardous pollutants from the environment or converting it into harmless compounds is crucial. Photocatalysis has been popular for degrading these pollutants into organic compounds (Zamri et al., 2021).  $\text{Ag}_2\text{CO}_3$  has been identified as an excellent photosensitizer and a highly active visible light-driven photocatalyst due to its small

band gap ( $\sim 2.17$  eV), which is beneficial for sunlight absorption (Rafaie et al., 2023). However, this photocatalyst displays photocorrosion behaviour that had hindered its functionality. Thus, many efforts were taken to improve such drawback. As previously reported, heterojunctions formation between Ag-based photocatalyst and  $\text{Nb}_2\text{O}_5$ , had shown promising properties, like enhanced separation of charge carriers (Osman et al., 2021). Thus, in this work we report the facile preparation of hybrid  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite with different weight ratios using facile chemical precipitation method, as well as its significant photodegradation performance on BPA.

## 2. METHODOLOGY

### 2.1. Synthesis of $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$ composite

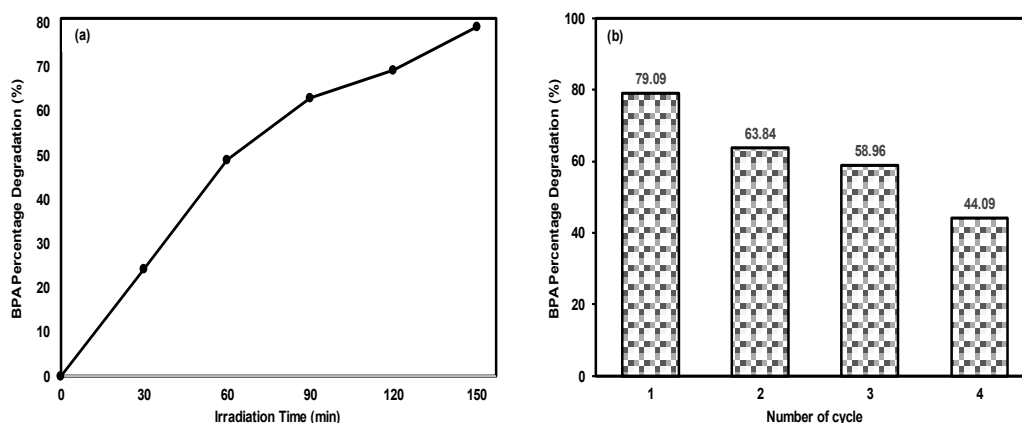
All chemicals and reagents were used as received and no purification step was done. The facile chemical precipitation method was employed to prepare the hybrid  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst at different molar ratios.  $\text{Nb}_2\text{O}_5$  (0.01 mol) was added into  $\text{AgNO}_3$  (0.02 mol) solution prior to sonication for 30 minutes.  $\text{Na}_2\text{CO}_3$  (0.01 mol) was added into the solution with constant stirring for 1 hour at room temperature. Subsequently, the precipitates of  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite were collected, washed several times with deionized water and dried at  $60^\circ\text{C}$  for 24 hours, yielding a pale-yellow powder named as Ag:Nb 1:1. The other molar ratios of  $\text{Ag}_2\text{CO}_3$  to  $\text{Nb}_2\text{O}_5$  (1:3, 1:5, 3:1 and 5:1) were synthesised likewise, meanwhile pure  $\text{Ag}_2\text{CO}_3$  were also be synthesized similarly but without  $\text{Nb}_2\text{O}_5$ .

### 2.2. Photocatalytic performance

The photocatalytic performance of  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite was evaluated by measuring the degradation of BPA. In all measurements, 0.1 g of photocatalyst were dispersed into BPA solution (200 mL, 10 mg/L) and magnetically stirred for 30 minutes in the dark to achieve the adsorption-desorption equilibrium prior to visible light irradiation for 150 minutes. At each time intervals, 5 mL of sample were extracted out and the change of concentration of BPA was monitored and analysed through UV-vis spectrophotometer at  $\lambda_{\text{max}} = 276$  nm.

## 3. RESULTS AND DISCUSSION

The photocatalytic activity of  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst was investigated through the photodegradation of BPA and the findings are as depicted in Figure 1 (a). It was found that within 150 minutes,  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst was degraded up to 79.09%. The stability and reusability of  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst was ascertained by the repetitive use of the photocatalyst under the same experimental conditions. After each cycle, the photocatalyst was recovered. Fresh BPA solution was used for each cycle without treating the photocatalyst to any kind of chemical or physical pretreatment before each cycle. As depicted in Figure 1 (b),  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst showed good stability trend even after four cycles. This result suggested that the  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst remained significantly photoactive with good stability towards visible light irradiation. The reduction in percentage degradation might be due to the photocorrosion of the photocatalyst as well as weight loss during the filtration process.



**Figure 1.** (a) The percentage degradation and (b) The reusability of  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst in degradation of BPA solution. Conditions:  $[\text{BPA}] = 10 \text{ mg/L}$ ;  $\text{pH} = 6.1$ ; 150 minutes irradiation time.

#### 4. CONCLUSION

In conclusion, it was revealed that  $\text{Ag}_2\text{CO}_3/\text{Nb}_2\text{O}_5$  composite photocatalyst exhibit a significant performance of BPA degradation under visible light irradiation in 150 minutes and can be used up to 4<sup>th</sup> cycle, where the reduction in photocatalytic performance was observed from 79.09% to 44.09%. To improve the photocatalytic performance of  $\text{Ag}_2\text{CO}_3$  based photocatalysts, another alternative such as adding transition metals e.g., Fe, Co, Ni and decorating the composite with noble metal nanoparticles e.g., Au, Ag also can enhance light absorption through surface plasmon resonance (SPR) effects. This study is hugely beneficial for its potential application in industrial wastewater treatment, especially towards the removal of EDCs, as well as dyes. Furthermore, this research is also a significant strategy for finding semiconductor photocatalyst that is useful towards cost-efficient wastewater treatment process due to the utilization of more affordable visible light irradiation compared to UV light irradiation that is costly.

#### ACKNOWLEDGEMENT

The authors gratefully acknowledge the financial support provided by the Ministry of Higher Education (MOHE) and Universiti Teknologi MARA (UiTM) for the YTR grant (600-RMC/YTR/5/3 (005/2020)), including the services and facilities provided to carry out the laboratory work in UiTM Pahang Branch, Jengka Campus and Centre for Functional Materials and Nanotechnology, Institute of Science, UiTM Shah Alam, Selangor.

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