

**UNIVERSITI TEKNOLOGI MARA**

**DIRECT DME PRODUCTION  
FROM CO<sub>2</sub> HYDROGENATION & MEOH  
DEHYDRATION OVER CZA & HZSM-5  
BIFUNCTIONAL CATALYST BED IN  
FIXED BED REACTOR**

**MUHAMMAD HAKIMI BIN KHAIRUDDIN**

Thesis submitted in fulfilment  
of the requirements for the degree of  
**Master of Science**  
**(Chemical Engineering)**

**Faculty of Chemical Engineering**

**February 2026**

## ABSTRACT

The direct synthesis of dimethyl ether (DME) from carbon dioxide (CO<sub>2</sub>) and hydrogen (H<sub>2</sub>) offers a promising carbon capture and utilization (CCU) route for clean fuel production. This study evaluates a bifunctional catalyst system comprising commercial CuO-ZnO-Al<sub>2</sub>O<sub>3</sub> (CZA) and HZSM-5 pellets in a fixed-bed reactor (FBR). Catalyst characterization was performed using XRD, FESEM, EDX, BET, mercury intrusion, TGA, H<sub>2</sub>-TPR, and NH<sub>3</sub>-TPD. CZA showed high hydrogen uptake (4900 μmol·g<sup>-1</sup>), confirming active Cu<sup>+</sup>/Cu<sup>0</sup> surface sites, while HZSM-5 exhibited moderate acidity with NH<sub>3</sub> desorption peaking near 217 °C. Pre-liminary tests compared catalyst form (powder vs pellet), packing structure (mixed vs sequential), and CZA:HZSM-5 mass ratios (1:1 to 4:1) under baseline conditions (200 °C, 40 bar, 500 ml g<sup>-1</sup> h<sup>-1</sup> GHSV). The optimal configuration—pellet form, sequential packing, 3:1 ratio—was selected for operating condition optimization. Temperature (150–250 °C), pressure (20–60 bar), and GHSV (250–583 ml g<sup>-1</sup> h<sup>-1</sup>) were varied using a one-factor-at-a-time approach. The best performance was achieved at 225 °C, 50 bar, and 500 ml g<sup>-1</sup> h<sup>-1</sup>, yielding 85.8% CO<sub>2</sub> conversion, 84.0% DME selectivity, and 97.9% DME yield. Carbon balance validation showed <0.5% deviation. A 12-hour stability test confirmed sustained activity and selectivity. Comparative analysis with literature benchmarks demonstrated superior performance. This study is the first to systematically optimize commercial CZA/HZSM-5 pellet configuration and operating conditions in a single-reactor setup, achieving the highest reported DME yield under non-synthesized catalyst conditions. These findings confirm the viability of commercial bifunctional pellets for scalable, energy-efficient direct DME synthesis.

## ACKNOWLEDGEMENT

I would like to express my sincere gratitude to several individuals and organizations who have supported me throughout my Master's research.

First and foremost, I extend my deepest appreciation to my supervisors, Ir. Dr. Muhammad Shafiq Mat Shayuti and Assoc. Prof. Ir. Ts. Dr. Nur Hashimah Alias, for their invaluable guidance, expertise, and encouragement throughout this research endeavour. Their insightful feedback and unwavering support have been instrumental to the successful completion of this thesis.

Also to thank Univesiti Teknologi MARA Faculty of Chemical Engineering, Oil & Gas Department and Institute of Postgraduate Studies for the support from staff together with the facilities provided to complete this study.

I am also grateful to Ir. Dr. Nur Hidayati Othman, Nor Hafizah Berahim and others from PETRONAS Research Sdn Bhd for providing access to resources, facilities, and valuable insights related to carbon utilization and relevancy of this research towards the oil and gas industry.

Additionally, I would like to acknowledge the Malaysia's Ministry of Higher Education for the MyBrain 2.0 scholarship, which provided crucial financial support for my studies.

Finally, I would like to express my gratitude to my parents, Khairuddin and \_\_\_\_\_, family, lab colleagues and friends for their unwavering support and understanding throughout this journey

# TABLE OF CONTENTS

	<b>Page</b>
<b>CONFIRMATION BY PANEL OF EXAMINERS</b>	<b>ii</b>
<b>AUTHOR'S DECLARATION</b>	<b>iii</b>
<b>ABSTRACT</b>	<b>iv</b>
<b>ACKNOWLEDGEMENT</b>	<b>v</b>
<b>TABLE OF CONTENTS</b>	<b>vi</b>
<b>LIST OF TABLES</b>	<b>ix</b>
<b>LIST OF FIGURES</b>	<b>xi</b>
<b>LIST OF SYMBOLS</b>	<b>xv</b>
<b>LIST OF ABBREVIATIONS</b>	<b>xvi</b>
<b>LIST OF NOMENCLATURE</b>	<b>xvii</b>
<b>CHAPTER 1 INTRODUCTION</b>	<b>1</b>
1.1 Research Background	1
1.2 Problem Statement	3
1.3 Research Objectives	4
1.4 Scope and Limitation	5
1.5 Significance of Study	6
<b>CHAPTER 2 LITERATURE REVIEW</b>	<b>7</b>
2.1 DME Properties and its Fuel Potential	7
2.2 Routes to Produce DME	9
2.2.1 Indirect DME Production	10
2.2.2 Direct DME Production	11
2.3 Catalyst Selection for Direct DME	13
2.3.1 CO <sub>2</sub> Hydrogenation Catalyst	13
2.3.2 MeOH Dehydration Catalyst	18
2.4 Important Characteristics of Bifunctional Catalyst and its Techniques	23
2.4.1 Crystalline Structure & Phase Composition	23
2.4.2 Morphology & Homogeneity	23

# CHAPTER 1

## INTRODUCTION

### 1.1 Research Background

Carbon dioxide (CO<sub>2</sub>) emissions resulting from human activities, particularly the burning of fossil fuels, deforestation, and industrial processes, have been identified as a significant driver of global warming and climate change. CO<sub>2</sub> contributes 76% of all greenhouse gas emissions as compared to other gasses including methane (CH<sub>4</sub>), nitrogen oxide (N<sub>2</sub>O), and fluorinated gases, making it the main driver of global climate change (Crippa et al., 2021). As the concentration of CO<sub>2</sub> in the atmosphere continues to rise, there is an urgent need for innovative solutions to mitigate these emissions.

Carbon capture and storage (CCS) technology has been proposed in industrial processes since 1920s as a method to reduce CO<sub>2</sub> emissions by capturing, transporting, and storing it in underground geological formations (IEA, 2016). Capturing, moving, and storing CO<sub>2</sub> in geological formations is considered as a viable way for lowering emissions of CO<sub>2</sub> into the atmosphere. However, CCS faces several challenges that limit its widespread implementation (Raza et al., 2019). The high cost of capturing, transporting, and storing CO<sub>2</sub> remains a major barrier. CCS is also energy-intensive, which reduces the efficiency of power plants and increases operational costs. CO<sub>2</sub> can also interact with storage reservoirs that can lead to geochemical and mechanical changes, threatening the long-term stability of storage sites. There is also a risk of CO<sub>2</sub> leakage, which could lead to environmental damage and safety concerns. While CCS shows potential for reducing emissions, the significant challenges associated makes this technology less attractive. As a result, increasing attention is being directed towards carbon capture and utilization (CCU).

CCU presents opportunities for converting CO<sub>2</sub> into valuable products, thereby offering both economic incentives and environmental benefits. CCU mitigates financial concerns by converting captured CO<sub>2</sub> into valuable products such as chemicals, building materials, and synthetic fuels, thereby reducing the overall cost of carbon capture while generating new revenue streams. This approach aligns with circular economy principles by promoting the reuse of waste materials, leading to a more resource-efficient and sustainable system. CO<sub>2</sub> can be transformed into a variety of chemical products,