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PROCEEDINGS OF JOHOR INTERNATIONAL INNOVATION INVENTION COMPETITION AND SYMPOSIUM 2024 (JIICaS 2024)



*“Flourish and Nurturing Sustainable
Innovation for a Prosperous Nation”*

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Preface

In the name of Allah, the Almighty who gives us the enlightenment, the truth, the knowledge and with regards to Prophet Muhammad (peace be upon him) for guiding us to the straight path. We thank to Allah for giving us guidance and strength to write this e-book.

This e-book compiles the extended abstracts that submitted to Johor International Innovation Invention Competition and Symposium 2024 (JIIICaS2024), where JIIICaS2024 is a virtual platform for all creative minds to share and present their invention and innovation. Each abstract gives a brief background on the innovation or project.

We hope that this e-book will help the readers to get to know the innovation done by the students and get some ideas to develop future innovation products.

Foreword Rector



Assalamualaikum warahmatullahi Wabarakatuh,
Salam Sejahtera, Salam Malaysia MADANI and
Salam UiTM Dihatiku.

In the name of Allah, the Most Gracious, the Most
Merciful.

It is a great honor to welcome you to the Johor
International Innovation, Invention, Competition, and
Symposium 2024 (JIICaS 2024). This event

connects various disciplines, focusing on education and engaging educators,
students, researchers, and innovators from all walks of life.

Innovation is not just about ideas; it demands perseverance, creativity, and
determination to turn those ideas into reality. The remarkable projects
showcased today highlight the dedication and spirit of all participants.
Initiatives like this not only explore new technologies but also cultivate skills
and leadership among our youth. At Universiti Teknologi MARA (UiTM) Johor
Branch, we are fully committed to fostering a dynamic culture of innovation,
promoting the commercialization of new products, and encouraging
meaningful collaborations with industry and society.

As we celebrate this event, I would like to extend my heartfelt gratitude to all
sponsors, judges, the College of Computing, Informatics and Mathematics,
UiTM Pasir Gudang Campus as the event organizer, as well as to the
researchers and participants for their hard work in making this event a
success. Let us continue striving for innovation and excellence. May the
ideas presented today inspire us and lay the groundwork for future
achievements.

Thank you.

Associate Professor Dr. Saunah Zainon
Rector
Universiti Teknologi MARA (UiTM)
Johor Branch

(A-ST063) ADVANCING SUSTAINABLE AND RAPID PRODUCTION OF HYDROGEN VIA NOVEL BOROPHENE CATALYST

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ABSTRACT

Hydrogen is a renewable energy resource with potential to replace fossil fuels and does not cause environmental concerns during energy generation. However, current mass-production methods involve fossil fuels, prompting urgent research into green hydrogen generation. The recent emergence of borophene oxide (BO), a material resembling graphene, has been theorized to enhance hydrogen production when used as a catalyst. Nevertheless, practical test in such application was not done. Thus, in this project, borophene was synthesized via modified Hummers' method and used to hydrolyze sodium borohydride, a reaction that produces hydrogen to proof the mentioned theory. Plate-like BO was successfully produced and characterized. More importantly, the experimental results showed that BO performed significantly better than TiO_2 , a common photocatalyst in where BO was able to produce 50 mL of H_2 in only 90 s. However, further doping the BO with Fe reduced the catalytic performance and took at least 3.33 times longer than BO to produce 50 mL of H_2 . Despite this, the project has successfully proven the catalytic properties of borophene in hydrogen production and borophene could be utilized in the future in the generation of sustainable energy.

Keywords: Borophene; Graphene; Hydrogen; Green Energy; Hydrogen Evolution Reaction

1.0 INTRODUCTION

Hydrogen (H_2) has emerged as an alternative to fossil fuels due to its exceptional properties of zero pollution emissions and high energy which has drawn considerable interest recently in energy utilization as a clean energy carrier. On 5 October 2023, the Malaysian government introduced the Hydrogen Economy Roadmap (HETR) to establish itself as a prominent hydrogen-based economy, generating revenues exceeding RM400 billion by 2050. Thus, this research may be able to contribute efforts to the Malaysian Government in HETR. Extensive research was done to produce renewable hydrogen which is environmentally friendly as it does not produce

greenhouse gas emissions during production. One such method proposed is the hydrolysis of hydrogen-rich compounds, such as sodium borohydride. Traditionally graphene (Gr) has been used in NaBH_4 hydrolysis, however, borophene, a crystalline atomic monolayer of boron, emerged as a new technology with greater catalytic activity potential than graphene in hydrogen evolution reaction which is theorised due to its outstanding properties which enable high catalytic activity from its extreme electron deficiency and abundant active sites.

2.0 OBJECTIVE

1. To synthesize BO and characterize its physicochemical properties.
2. To evaluate the effects of various process parameters (reaction time, catalyst and reactant amounts) on H_2 production.
3. To compare the kinetic profiles of H_2 production induced by various commonly used catalysts alongside borophene.

3.0 METHODOLOGY

Borophene was produced using a modified Hummer's method. Borophene itself is highly soluble in water which is attributed by its hydrophilic properties and being a nanocatalyst which has a high surface ratio over volume ratio, contributing its ability as a catalyst.

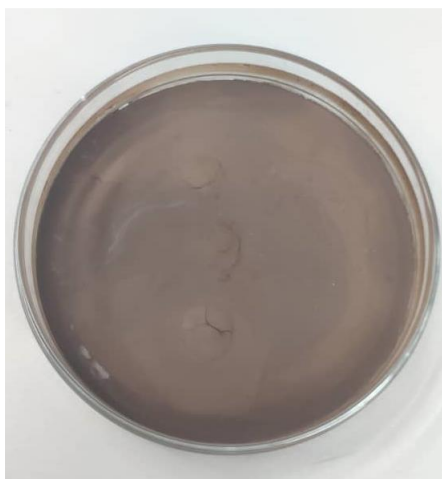


Figure 1: Synthesized dried BO.



Figure 2: Setup of NaBH₄ hydrolysis.

4.0 RESULTS

This study showed that the synthesis of BO as a novel catalyst for the purpose of H₂ production via NaBH₄ hydrolysis. Research on the comparison between nanomaterials (as catalysts) is scarce and limited, which makes this study necessary. Here, the H₂ production of NaBH₄ hydrolysis was carried out with BO and TiO₂.

5.0 CONCLUSION

The experimental results showed that BO performed significantly better than TiO₂, a common photocatalyst in where BO was able to produce 50 mL of H₂ in only 90 s. However, further doping the BO with Fe reduced the catalytic performance and took at least 3.33 times longer than BO to produce 50 mL of H₂. Despite this, the project has successfully proven the catalytic properties of borophene in hydrogen production and borophene could be utilized in the future in the generation of sustainable energy