

SUBMISSION FOR EVALUATION FINAL YEAR PROJECT 2 - RESEARCH PROJECT

IRON DECORATED BIOCHAR FROM MANGO KERNEL FOR THE ADSORPTION OF PHENOL

Name : MUMTAZAH FAQIHAH BINTI GHALIP

Student ID : 2023115761

Program : APPLIED CHEMISTRY

Course code : FSG671

Mobile Phone :

E-mail : mumtazahfg02@gmail.com

Approval by Main Supervisor:

I certify that the work conducted by the above student is completed and approve this report to be submitted for evaluation.

Supervisor's name : DR. JEYASHELLY ANDAS

Date : 29TH JULY 2025

Turnitin Similarity % : 4%

Signature :

IRON DECORATED BIOCHAR FROM MANGO KERNEL FOR THE ADSORPTION OF PHENOL

MUMTAZAH FAQIHAH BINTI GHALIP

Final Year Project Submitted in
Partial Fulfilment of the Requirements for the
Degree of Bachelor of Science (Hons.) Applied Chemistry
In the Faculty of Applied Sciences,
Universiti Teknologi MARA

AUGUST 2025

ABSTRACT

IRON DECORATED BIOCHAR FROM MANGO KERNEL FOR THE ADSOPRTION OF PHENOL

Phenol contamination in water poses serious environmental and health risks, necessitating the development of sustainable and cost-effective treatment methods. This study investigates the development of mango kernel biochar (MKB) that was going through pyrolysis process, aimed at enhancing the adsorption of phenol from aqueous solutions. MKB was synthesized and subsequently modified with iron (Fe-MKB) through an impregnation process, and both materials were thoroughly characterized using FTIR, XRD, SEM-EDX, CHNS, AAS, and UV-Vis spectroscopy. The successful incorporation of iron into the biochar matrix was confirmed through increased iron content and the appearance of iron-related features in both elemental and structural analyses. Although the iodine number of Fe-MKB indicated some reduction in microporosity due to partial pore blockage, morphological analysis showed a more developed pore structure. Importantly, Fe-MKB demonstrated superior phenol adsorption performance compared to unmodified MKB, with the capacity increasing from 18.70 mg/g to 35.00 mg/g. This enhancement is attributed to improved surface interaction and catalytic properties introduced by the iron species. The findings suggest that Fe-MKB offers a sustainable and efficient solution for removing phenol from water, with potential for broader environmental remediation applications.

TABLE OF CONTENT

TABLE OF CONTENT	i
LIST OF TABLES	iii
LIST OF FIGURES	iv
LIST OF SYMBOLS	v
LIST OF ABBREVIATIONS	vii
CHAPTER 1 INTRODUCTION	
1.1 Research Background	1
1.2 Problem Statement	4
1.3 Research Questions	4
1.4 Objectives	5
1.5 Significance of Study	5
CHAPTER 2 LITERATURE REVIEW	
2.1 Introduction to Biochar	6
2.1.1 Characteristics of Biochar	7
2.1.2 Sources of Biochar	8
2.1.3 Pyrolysis Method of Biochar	9
2.1.4 Iron Impregnation Method for Biochar Decoration	10
2.1.5 Application of Biochar	12
2.2 Mango	14
2.2.1 Mango Kernel	15
2.3 Iron	17
2.3.1 Applications of Iron Decorated Biochar	18
2.4 Phenol	19
2.4.1 Environmental and Health Concerns	20
2.5 Adsorption Mechanism for Phenol Removal	22
2.5.1 Biochar as an Adsorbent	22
2.5.2 Adsorption of Phenolic Compounds	23
CHAPTER 3 RESEARCH METHODOLOGY	
3.1 Raw Materials	27
3.2 Analytical Equipments	27
3.3 Synthesize of Biochar from Mango Kernel	27

3.4 Biochar Decorated with Iron	
3.5 Characterization Methods	28
3.5.1 Fourier Transform Infrared Spectroscopy (FTIR)	28
3.5.2 X-ray Diffraction (XRD)	29
3.5.3 Atomic Absorption Spectroscopy (AAS)	29
3.5.4 Scanning Electron Microscopy with Energy-Dispersive X-ray Spectroscopy (SEM/EDX) and Elemental Mapping	30
3.5.5 Proximate and Ultimate Analysis	30
3.6 Iodine Number Determination	31
3.7 Evaluation of Phenol Adsorption Efficiency	32
3.8 Experimental Designs/Flow Chart	33
CHAPTER 4 RESULTS AND DISCUSSIONS	
4.1 Yield of Biochar (MKB) and Iron-Decorated Biochar (Fe-MKB)	34
4.2 Characterization Results	35
4.2.1 Fourier Transform Infrared Spectroscopy (FTIR)	35
4.2.2 X-ray Diffraction Analysis (XRD) Analysis	37
4.2.3 Scanning Electron Microscopy with Energy-Dispersive X-ray Spectroscopy (SEM/EDX) and Elemental Mapping	39
4.2.4 Elemental Analysis (CHNS)	42
4.2.5 Atomic Absorption Analysis (AAS)	44
4.2.6 Proximate Analysis	44
4.2.7 Iodine Number Determination	47
4.3 Phenol Adsorption Efficiency via UV-Visible Spectroscopy	48
CHAPTER 5 CONCLUSION AND RECOMMENDATIONS	
5.1 Conclusion	50
5.2 Recommendations	50
CITED REFERENCES	52
APPENDICES	67
CURRICULUM VITAE	73