UNIVERSITI TEKNOLOGI MARA

DECOLOURIZATION OF TEXTILE EFFLUENT BY ADSORPTION ON NAOH TREATED TEXTILE VENUS CLAM (PAPHIA TEXTILE)

MUHAMMAD RAZIF BIN SAIDINA OMAR

Thesis submitted in fulfillment of the requirements for the degree of Bachelor of Engineering (Hons.) Chemical and Process

Faculty of Chemical Engineering

January 2018

ACKNOWLEDGEMENT

First of all, I would like to say thank you to my supervisor Syed Shatir Asghrar Syed Hassan for being a very good guider for me while I was doing the research project. She had given me appropriate examples and knowledges in order to make me understand more about my research project. She also makes sure that I understand everything she told me and explain while doing this project. Besides that, I would like to state my gratitude to all of my lectures for the information and encouragement that they put into me. They give me a lot of advice about my study and give me some idea on how to manage my research project. Lastly, I would like to express my gratitude to my fellow friends and my parents who tried their best to give support and encouragement to keep up with the task given or by supporting financially and emotionally throughout my degree.

ABSTRACT

In this work the feasibility of employing Textile Venus Clam (TVC) to remove Basic Red 46 (BR46), an azo dye from its aqueous solutions was investigated. The parameters that influence the adsorption process such as adsorbent dosage, initial dye concentration and contact time were studied in batch experiments. The adsorption of Basic Red 46 was took place at neutral pH of 7 and room temperature. Furthermore, the adsorbent was characterized by using Brunaeur-Emmett Teller (BET) to study the surface area of the adsorbent. The TVC was washed using deionized water and treated with NaOH solution. Then the adsorbent was dried for 24 hours at 100°C. The adsorbent then crushed and calcined at 400, 500 and 600°C for 3 hours and stored. For the effect of adsorbent dosage, a range of dosage from 0.3-0.9g of adsorbent in 100mL of 20 ppm of BR46 solution. The process was carried out for 60 minutes in an orbital shaker at 200 rpm. For the effect of initial dye concentration, the optimal dosage was used and a range of initial concentration from 20-60 ppm of BR46 solution. The solution was shake for the same time and speed. The samples were filtered by using filter paper and the dye concentration was measured using UV-Vis Spectrophotometer. The BET analysis showed that the adsorbent has BET surface area of 2.4758 m2/g and pore volume of 0.003754 cm3/g. The highest adsorption of BR46 was at 0.9g adsorbent dosage which is 87.48% removal. The highest removal for initial concentration of 20 ppm concentration of dye was 87.6% removal. Thus, it was concluded that TVC can be used effectively as an adsorbent fort the removal of BR46 from aqueous solution.

TABLE OF CONTENTS

CONTENTS

AUTHOR'S DECLARATION
SUPERVISOR'S CERTIFICATION iii
COORDINATOR'S CERTIFICATIONiv
ACKNOWLEDGEMENT
ABSTRACTvi
TABLE OF CONTENTSvii
LIST OF TABLESix
LIST OF FIGURES
LIST OF ABBREVIATION xiii
CHAPTER 1
INTRODUCTION
1.1 RESEARCH BACKGROUND1
1.2 PROBLEM STATEMENT
1.3 OBJECTIVES
1.4 SCOPE OF RESEARCH
CHAPTER 2
LITERATURE REVIEW
2.1 METHODS OF DYE REMOVAL
2.1.1 Nanofiltration
2.1.2 Adsorption
2.1.3 Fungal decolourization7
2.2 CATEGORIES OF NATURAL WASTES
2.2.1 Agricultural wastes
2.2.2 Plant wastes10
2.2.3 Biomass wastes
2.2.4 Fruit wastes14
2.3 CLASSES OF DYE16
2.3.1 Acid dyes (Anionic)16
2.3.2 Basic dyes (Cationic)17

CHAPTER 1

INTRODUCTION

1.1 RESEARCH BACKGROUND

The world nowadays is having a big environmental problem such as climate change, pollutions, land degradation, nuclear issues and others. These problems will not only affect our life style but also the future generations. Like colouring agents or dye, they had been polluting the environment because of the increasing demand in industries and had become a serious concern (Rafatullah, Sulaiman, Hashim, & Ahmad, 2010). There are many types of dye that are used by the industries as their colouring agents (Anastopoulos & Kyzas, 2014). Dyes like Methyl Orange, Methyl Red, Malachite Green and Methylene Blue are major pollutants of fresh water reserves because of their toxic and carcinogenic properties (Mane & Babu, 2011; Mane & Vijay Babu, 2013; Yagub, Sen, Afroze, & Ang, 2014).

The discharge of dues into the receiving water causes a big concern from the environmental point of view. Several industries that are polluting the fresh water with dyes such as textile industry, batik industry, paint industry and printing ink production (Rashidi, Sulaiman, & Hashim, 2012; Wang, Zhu, Jiang, Hu, & Shen, 2016). Damage had done to the environment because of the failure in treating the wastewater whether it is from the industrial and domestic. It is also affecting the human health and cause problems such as organ damage, reduced growth and development, nervous system failure oxidative stress (Dahri, Kooh, & Lim, 2015; Subramaniam & Kumar Ponnusamy, 2015). That is why it is very important to remove or reduce the dyes before it is discharge into the environment.