

# **EFFECT OF IODINE INCORPORATION TO THE ELECTRICAL PROPERTIES OF AMORPHOUS CARBON THIN FILMS**

This thesis is presented in partial fulfillment for the award of the

**Bachelor of Engineering (Hons) Electronic**

**Universiti Teknologi MARA**

**(JULY 2013)**



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## ACKNOWLEDGEMENT



With my humbleness and high gratitude to Allah S.W.T for give me the ideas, strength and patience that enables me to complete this Final Year Project Thesis. I wish to express my heartily gratitude to my supervisor Prof. Engr. Dr. Mohamad Rusop Mahmood for the guidance, attention and enthusiasm given for realization of this project. I'm also would like to acknowledge Master student Miss Dayana Binti Kamaruzaman that given me kind guidance, constructive suggestion and continous encouragement in a very supportive manner during preparation of this final year project. Last but not least, special of thanks dedicated to my family, all the seniors and staff at the NANO-ElecTronic Centre (NET) Faculty of Electrical Engineering and NANO-Tech Centre (NST) Institute of Science and also all members from Faculty of Electrical Engineering batch 2010-2013 and all lecturers who gave me advice during development of this project. Thank you and may Allah bless you.

## ABSTRACT

Amorphous carbon (a-C) thin films were deposited on glass and silicon substrates by thermal chemical vapor deposition (CVD) technique using camphor oil as the precursor. After deposited the a-C thin films, the iodine (I) was doped on the a-C thin films using the same technique of deposition of a-C thin film. All the samples were grown in fixed conditions except the doping temperature parameter was varied. The effect of doping temperature in the a-C and a-C:I thin films on electrical, structural and optical properties was characterized by using a standard two-probe method using BUKOH KEIKI (CEP-2000) Solar simulator/Spectral sensitivity Measurement, RAMAN spectroscopy and UV-Vis-NIR spectroscopy respectively. The conductivity of a-C:I thin films increased to  $7.44 \times 10^{-3} \text{ S.cm}^{-1}$  with the doping temperature up to  $450^{\circ}\text{C}$  and it shows photoresponse is 1.063 at doping temperature  $450^{\circ}\text{C}$ . The UV-Vis-NIR analysis was used to obtain the optical absorption coefficient and optical band gap. The absorption coefficient at sample  $450^{\circ}\text{C}$  increase from  $8763 \text{ cm}^{-1}$  to  $39292 \text{ cm}^{-1}$  and the optical band gap decrease from 0.33eV to 0.08eV. The RAMAN scattering analysis was used to prove the amorphous structure of a-C and a-C:I thin films and it revealed at D-peak ( $1340 \text{ cm}^{-1}$ ) and G-peak ( $1590 \text{ cm}^{-1}$ ).

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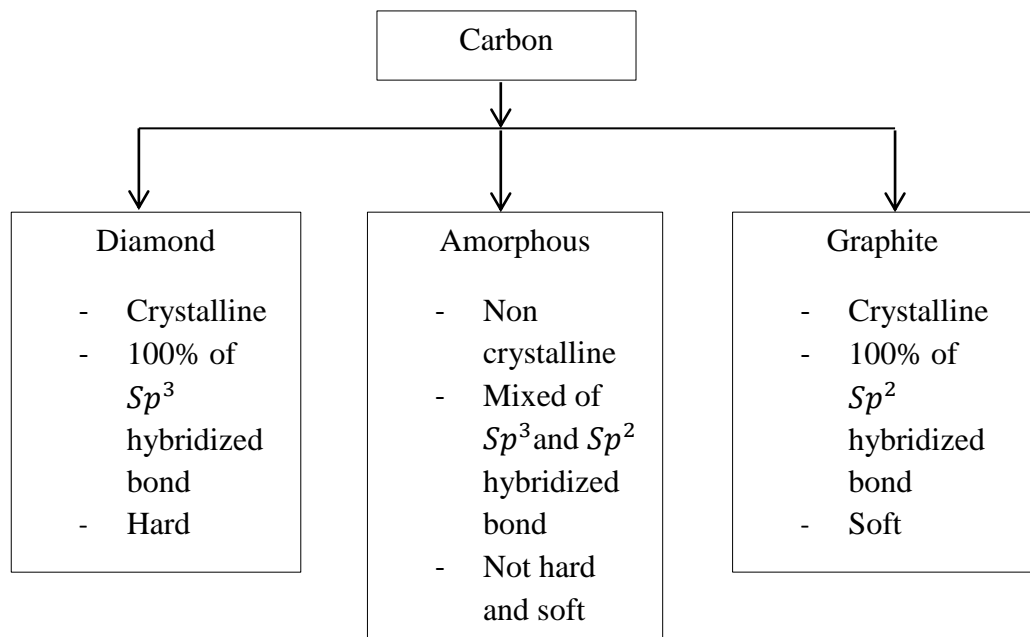
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# CHAPTER 1

## INTRODUCTION

### 1.1 BACKGROUND OF STUDY

Carbon is the first element in group 14 in periodic table. Carbon is an attractive material in a variety of stable forms such as graphite, diamond, nanotubes and fullerenes [1]. Figure 1.1 shows the type of carbon.



**Figure 1.1: Type of Carbon**